# **Supporting Information**

# **Electronic Alteration on Oligothiophenes by o-Carborane: Electron Acceptor Character of o-Carborane in Oligothiophene Frameworks with Dicyano-Vinyl End-On Group**

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## NMR Spectra

Compound 5



Compound 6



Compound 1a





Compound 7

## <sup>1</sup>H NMR





Compound 9b



Compound 10b



Compound **1b** 





Compound 9c



Compound **10c** 



Compound 1c

#### $^{1}$ H NMR





Compound 11



Compound 2a

#### <sup>1</sup>H NMR



i.



Compound 12b

<sup>1</sup>H NMR



<sup>13</sup>C NMR



S17

1

J

Compound 13b

## <sup>1</sup>H NMR



## <sup>13</sup>C NMR



S18

Compound 2b

#### <sup>1</sup>H NMR



## <sup>13</sup>C NMR



S19

Compound 12c





Compound 13c

#### <sup>1</sup>H NMR





Compound 2c



ppm

S22

ı.



Figure S1. ORTEP drawing of compound 1a with 30% probability for the thermal ellipsoids. Hydrogens were omitted for clarity.



**Figure S2.** ORTEP drawing of compound **2a** with 30% probability for the thermal ellipsoids. Hydrogens were omitted for clarity.

Identification code	1a	2a
Empirical formula	$C_{10}H_{14}B_{10}N_2S$	$C_{18}H_{16}B_{10}N_4S_2$
Formula weight	302.39	460.57
Temperature	293(2) K	293(2) K
Wavelength	0.71073 Å	0.71073 Å
Crystal system, space group	monoclinic, $P2_1/c$	Monoclinic, C 2/c
Unit cell dimensions	<i>a</i> = 18.348(6) Å	a = 15.986(8) Å
	$b = 7.522(2) \text{ Å} \qquad \beta = 104.643(6)^{\circ}$	$b = 11.333(6)$ Å $\beta = 114.529(8)$ °
	c = 12.006(4) Å	c = 14.165(7) Å
Volume	1603.0(9) Å <sup>3</sup>	2334(2) Å <sup>3</sup>
Z, Calculated density	4, 1.253 Mg/m <sup>3</sup>	4, 1.310 Mg/m <sup>3</sup>
Absorption coefficient, $\mu$	0.190 mm <sup>-1</sup>	0.244 mm <sup>-1</sup>
<i>F</i> (000)	616	936
Crystal size	$0.23 \times 0.15 \times 0.09 \text{ mm}$	$0.15 \times 0.13 \times 0.11 \text{ mm}$
$\theta$ range for data collection	1.15 to 28.40°	2.278 to 28.850°
Limiting indices	-24≤ <i>h</i> ≤24, -10≤ <i>k</i> ≤9, -16≤ <i>l</i> ≤16	-21≤ <i>h</i> ≤21, -14≤ <i>k</i> ≤15, -18≤ <i>l</i> ≤18
Reflections collected / unique	20624 / 3984 [ <i>R</i> <sub>int</sub> = 0.0871]	8144 / 2567 [ <i>R</i> <sub>int</sub> = 0.0440]
Completeness to $\theta = 28.40$	99.2 %	85.6 %
Refinement method	Full-matrix least-squares on $F^2$	Full-matrix least-squares on $F^2$
Data / restraints / parameters	3984 / 0 / 208	2567 / 0 / 155
Goodness-of-fit on $F^2$	1.066	1.184
Final <i>R</i> indices $[I > 2\sigma(I)]$	$R_1 = 0.0758, wR_2 = 0.2002$	$R_1 = 0.0932, wR_2 = 0.2462$
<i>R</i> indices (all data)	$R_1 = 0.1278, wR_2 = 0.2479$	$R_1 = 0.1313, wR_2 = 0.2616$
Largest diff. peak and hole	0.726 and -0.411 e. Å <sup>3</sup>	0.504 and -0.578 e. Å $^3$

Table S1. Crystal data and structure refinement for 1a and 2a

 $\overline{{}^{a}R_{1} = \sum ||F_{o}| - |F_{c}||} \text{ (based on reflections with } F_{o}^{2} > 2\sigma F^{2}\text{)}, \ {}^{b}wR_{2} = [\sum [w(F_{o}^{2} - F_{c}^{2})^{2}]/\sum [w(F_{o}^{2})^{2}]]^{1/2}; \ w = 1/[\sigma^{2} + (F_{o}^{2})^{2}]/\sum [w(F_{o}^{2})^{2}]^{1/2}; \ w = 1/[\sigma^{2} + (F_{o}^{2})^{2}]/\sum [w(F_{o}^{2})^{2}]/\sum [w(F_$ 



Figure S3. Absorption and emission spectra of 1 in dichloromethane (DCM).



Figure S4. Absorption and emission spectra of 2 in DCM.



**Figure S5.** CVs of **DCVT1**, **DTC**, **1** and **2** ( $10^{-3}$  M substrate) in 0.10 M Bu<sub>4</sub>NPF<sub>6</sub> in DCM with scan rate 100 mV s<sup>-1</sup>.



Figure S6. Mataga–Lippert plots for CT emissions of 1b (black), 1c (blue) and 2c (red).

**DFT Calculations.** The ground-state geometries of **1a–1c** and **2a–2c** were optimized at the density function theory (DFT) level. The characterization of the low-lying excited singlet states relies on the time-dependent DFT (TD-DFT) calculation that is performed on the basis of the ground-state geometry by B3LYP<sup>S1</sup> density functional theory (DFT), using a 6-31G(d,p)<sup>S2</sup> basis set and no imaginary frequencies were found. All calculations were performed with the Gaussian 09 package.<sup>S3</sup> Molecular orbital plots were made by using Chem3D Pro (version 10.0).

	Energy (cm <sup>-1</sup> )	Wavelength (nm)	Osc. Strength	Orbital contributions
	29540.26	338.5	0.779	HOMO $\rightarrow$ LUMO (98%) and H-1 $\rightarrow$ LUMO (2%)
1a	34270.73	291.8	0.036	H-1 $\rightarrow$ LUMO (94%), HOMO $\rightarrow$ LUMO (2%), and HOMO $\rightarrow$ L+1 (2%)
	46613.52	214.5	0.0429	H-12 $\rightarrow$ LUMO (15%), H-11 $\rightarrow$ LUMO (75%), and H-9 $\rightarrow$ L+2 (2%)
2a	25494.56	392.2	0.1396	$H-1 \rightarrow L+1 (12\%)$ and HOMO $\rightarrow$ LUMO (88%)
	28776.45	347.5	0.4143	H-1 $\rightarrow$ L+1 (85%), HOMO $\rightarrow$ LUMO (11%), and H-2 $\rightarrow$ LUMO (3%)
	29900.79	334.4	0.8843	H-1 → LUMO (41%), HOMO → L+1 (56%), and H-2 → L+1 (2%)
	33844.87	295.5	0.0221	$H-2 \rightarrow L+1 (96\%)$

Table	S2.	Selected	energy	levels,	orbital	contributions,	and	oscillation	strengths	of <b>1</b>	a and 2a	a
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## CARTESIAN COORDINATES AND ENERGIES FOR OPTIMIZED STRUCTURES

#### **1a**

Energy: -1145.80102527 a.u.

S	-1.1207	0.3437	0.0434
Ν	-7.0722	-0.0944	-0.0520
Ν	-3.6080	2.6505	0.0340
В	2.1793	0.5909	-1.3566
Η	1.4142	0.6613	-2.2512
В	1.9880	1.4357	0.1990
Η	1.0798	2.1830	0.3198
В	2.3922	0.3085	1.5224
Η	1.7398	0.2977	2.5078
В	2.8342	-1.2529	0.7957
Η	2.4833	-2.3015	1.2055
В	4.2810	-1.0154	-0.2051
Η	4.9543	-1.9630	-0.4223
В	3.8843	0.1181	-1.5209
Н	4.2912	-0.0781	-2.6136
В	3.4420	1.6826	-0.7966
Η	3.6262	2.7040	-1.3668
В	3.5811	1.5122	0.9803
Η	3.8656	2.4271	1.6768
В	4.1001	-0.1601	1.3469
Н	4.7526	-0.4522	2.2912
В	4.7553	0.6950	-0.0829
Н	5.8934	1.0157	-0.1532
С	1.6274	-0.2320	0.0795
С	2.7342	-0.9677	-0.8987
Н	2.3478	-1.7862	-1.4921
С	0.2308	-0.7527	0.0540
С	-0.1782	-2.0700	0.0273
Н	0.5059	-2.9089	0.0418
С	-1.5804	-2.2057	-0.0036
Η	-2.0956	-3.1594	-0.0248
С	-2.2566	-0.9933	0.0041
С	-3.6844	-0.8894	-0.0134
Н	-4.1958	-1.8486	-0.0351
С	-4.4951	0.2132	-0.0093
С	-5.9175	0.0481	-0.0324
С	-4.0064	1.5568	0.0153



**1b** Energy: -1697.62291184 a.u.

S	0.9092	-1.0383	-0.0848
В	4.1934	-0.6792	1.2712
Н	3.4697	-1.0203	2.1377
В	4.1857	-1.3663	-0.3710
Н	3.4690	-2.2815	-0.5896
В	4.3220	-0.0395	-1.5550
Н	3.6810	-0.0654	-2.5477
В	4.4035	1.4875	-0.6508
Н	3.8292	2.4727	-0.9502
В	5.8735	1.4702	0.3455
Н	6.3194	2.5141	0.6773
В	5.7517	0.1373	1.5201
Н	6.1113	0.2949	2.6355
В	5.6640	-1.3948	0.6194
Н	6.0745	-2.4074	1.0769
В	5.7521	-1.0004	-1.1252
Н	6.2307	-1.7466	-1.9114
В	5.8814	0.7751	-1.2942
Н	6.4476	1.3094	-2.1870
В	6.7196	-0.0663	0.0445
Н	7.9012	-0.1308	0.1009
С	3.4590	0.1521	-0.0779
С	4.3821	1.0012	1.0005
Н	3.8246	1.6419	1.6714
С	1.9825	0.3373	-0.0249
С	1.2727	1.5112	0.0763
Н	1.7398	2.4873	0.1156
С	-0.1294	1.3190	0.1129
Н	-0.8403	2.1331	0.1957
С	-0.5035	-0.0080	0.0332
С	-1.8329	-0.5738	0.0302
S	-3.2393	0.4631	-0.0092
С	-2.1988	-1.9128	0.0558
С	-4.3226	-0.9219	0.0181
Н	-1.4803	-2.7234	0.0870
С	-3.5888	-2.1043	0.0481
С	-5.7469	-0.8758	0.0098
Н	-4.0651	-3.0782	0.0681
Н	-6.2208	-1.8543	0.0238
С	-6.6042	0.1957	-0.0125
С	-8.0177	-0.0266	-0.0159
С	-6.1688	1.5564	-0.0316
Ν	-9.1663	-0.2158	-0.0184
Ν	-5.8111	2.6647	-0.0464



**1c** Energy: -2249.4244680 a.u.

a	0 (050	0 7024	0 1 407
5	-2.6250	0.7234	-0.1407
В	-5.8790	1.0243	1.2104
Η	-5.0786	1.2995	2.0317
В	-5.8150	1.5804	-0.4796
Н	-4 9632	2 3415	-0.7865
R	-6 2187	0.2163	-1 5536
н	-5.6216	0.0596	-2 5614
R	6 5210	1 2042	0 5305
ы П	6 1275	22042	0.7671
П	-0.13/3	-2.2931	-0.7071
D	-7.9508	-0.8003	0.4909
H D	-8.5330	-1./928	0.91/9
В	-7.5425	0.5075	1.5603
H	-/.8811	0.4958	2.6933
В	-7.2297	1.9324	0.5426
Н	-7.4462	3.0307	0.9303
В	-7.4503	1.4349	-1.1636
Н	-7.8255	2.1934	-1.9931
В	-7.8818	-0.2999	-1.1912
Н	-8.5627	-0.7936	-2.0256
В	-8.5167	0.7668	0.0984
Н	-9.6676	1.0367	0.1781
С	-5.3438	-0.0146	-0.0886
С	-6.3594	-0.6130	1.0763
Н	-5.8908	-1.2904	1.7784
С	-3.9199	-0.4477	-0.0506
C	-3 4189	-1 7220	0.0713
H	-4 0454	-2.6030	0.1342
C	-2 0035	-1 7708	0.1012
с ц	1 1/10	2 6024	0.1027
II C	1 4070	0 5209	0.2030
C	-1.40/9	-0.3298	-0.0009
C C	-0.0013	-0.19/0	-0.0074
2	1.21/8	-1.4301	-0.0440
C	0.5954	1.0503	0.01/8
C	2.5212	-0.2785	-0.0127
H	0.0297	1.9/44	0.0483
C	2.0053	1.0050	0.0139
Н	2.6291	1.8914	0.0369
С	3.8980	-0.6987	-0.0140
S	5.1912	0.4798	-0.0022
С	4.4041	-1.9939	-0.0218
С	6.4134	-0.7863	-0.0057
Н	3.7740	-2.8755	-0.0286
С	5.8048	-2.0391	-0.0172
С	7.8235	-0.5942	0.0032
Н	6.3802	-2.9583	-0.0210
Н	8.3951	-1.5193	-0.0018
С	8.5685	0.5599	0.0178
С	9.9968	0.4808	0.0255
С	7.9979	1.8694	0.0268



11.1586 0.4070 0.0319 7.5308 2.9364 0.0344 Ν

Ν

2a Energy: -1959.37135122 a.u.

С	2.2550	-0.7297	0.4927
С	2.2550	0.7298	-0.4927
В	3.5985	-1.7007	0.0817
В	2.6950	0.8070	1.1682
В	2.6950	-0.8069	-1.1681
В	4.4609	0.8189	1.1943
В	3.5986	1.7008	-0.0816
В	3.5889	-0.6888	1.5476
В	5.0273	-0.7316	0.5031
В	4.4609	-0.8188	-1.1943
В	3.5890	0.6889	-1.5476
В	5.0273	0.7316	-0.5031
Н	3.4386	-2.8698	0.1460
Н	1.9370	-1.3137	-1.9158
Н	3.4389	1.1700	-2.6165
Н	5.0364	-1.3999	-2.0506
Н	6.0197	1.2635	-0.8707
Н	6 0197	-1 2635	0 8707
Н	3.4388	-1.1700	2.6165
Н	1 9370	1 3138	1 9158
Н	3 4 3 8 7	2 8698	-0 1460
н	5.0364	1 3999	2 0506
C	0.9551	1 2987	-0.9254
C	0.2016	1.2087	-2 1872
S	-0.0391	2 2459	0 1469
C	-0.0371	1 9000	-2 2002
н	0.868/	0.6676	-2.2772
C II	1 2172	2 5200	1 1 2 1 6
ч	1 /06/	1 0/7/	3 2004
n C	0.0551	1.9474	0.0254
C	0.9331	1 2086	0.9234
c c	0.4010	-1.2080	2.1072
с С	-0.0391	1 2002	2 2001
	-0.0195	-1.0990	2.2991
п	0.0005	-0.0074	2.9990
С u	-1.21/2	-2.5209	2 2002
п	-1.4003	-1.9472	0.0060
C	-2.4514	3.2097	-0.9900
	-2.9792	5.8929 2.2447	0.0920
н С	-5.0021	3.3447	-1.9183
C	-2.4314	-3.2097	0.9959
U U	-2.9791	-3.8930	-0.0920
Н	-3.0021	-3.3440	1.9183
	-4.2237	-4.5895	0.0303
N	-5.2355	-5.1555	0.1459
	-2.3812	-5.88/1	-1.3916
IN C	-1.8953	-3.8/40	-2.4496
	-2.3813	3.8869	1.391/
IN C	-1.8955	5.8/38	2.4496
C	-4.2238	4.5894	-0.0363



N -5.2356 5.1534 -0.1459

**2b** Energy: -3063.03110368 a.u.

С	2.9709	2.9270	-0.1883
С	3.8834	1.4865	0.3832
B	3.5740	4.3126	0.5936
B	4.1443	2.1380	-1.1858
B	3.4232	2.7600	1.4620
B	5 6108	3 1210	-1 0591
B	5 4955	1 6038	-0 1484
B	4 0232	3 9167	-1 0820
B	5 2641	4 4667	0.0683
B	4 8723	3 7544	1 6646
B	5 0338	1 9915	1.5262
B	6 1604	3 0478	0.6437
Н	2 7862	5 1 5 9 6	0.8369
Н	2.7002	2 5396	2 2476
Н	5 2604	1 2457	2.2470
Н	5.0822	4 3000	2.4140
н Н	7 3054	3 0877	0.9454
н Н	5 7568	5 5377	-0.0486
н Н	3 5 5 3 9	J.JJ77	-0.0480
и П	3.5555	1 5170	2 1124
и П	6.0314	0.5861	-2.1124 0.4210
П Ц	6 3 4 6 5	2 2168	-0.4210
II C	0.5405 3 1601	0.1003	-1.9620
C C	2 6 4 0 1	0.1995	0.5150
C C	2.0491	-0.3492	0.8272
S C	2.9010	-0.0093	-0.6372
	2.0920	-1.0343	1.4790
п	2.0809	0.1330	2.0237
	2.1820	-2.0840	0.1730
п	1.0303	-2.2203	2.2798
C	1.5498	2.7033	-0.5000
C	1.0528	2.0139	-1.8282
<b>S</b>	0.2682	2.8223	0.6283
C II	-0.3/88	2.5510	-1.8574
H	1.6541	2.5698	-2./135
C U	-0.9590	2.0520	-0.6067
H	-0.9534	2.4517	-2.//11
C	-2.3606	2.6269	-0.2559
C	-2.9510	2.9477	0.9599
5	-3.5/12	2.1490	-1.4141
C	-4.3501	2.8213	0.9650
H	-2.3816	3.2815	1.8196
C	-4.8/15	2.3989	-0.2537
H	-4.9648	3.0384	1.8284
C	1.7274	-3.3400	-0.3777
C	2.0447	-3.9056	-1.6049
S	0.6376	-4.3636	0.5266
C	1.4293	-5.1499	-1.8126
H	2.7170	-3.4398	-2.3157
С	0.6284	-5.5682	-0.7539



Н	1.5629	-5.7494	-2.7063
С	-6.2114	2.1495	-0.6849
С	-7.3825	2.2580	0.0234
Н	-6.3288	1.8277	-1.7164
С	-0.0849	-6.8029	-0.7436
С	-0.8985	-7.3534	0.2140
Н	0.0448	-7.3956	-1.6459
С	-1.5072	-8.6274	-0.0190
Ν	-1.9984	-9.6640	-0.2166
С	-1.1868	-6.7226	1.4634
Ν	-1.4195	-6.2047	2.4802
С	-7.4463	2.6653	1.3913
Ν	-7.4823	2.9997	2.5061
С	-8.6270	1.9549	-0.6145
Ν	-9.6358	1.7075	-1.1398

**2c** Energy: -4166.68213352 a.u.

С	0.8104	5.7030	0.4044	
С	-0.8090	5.7030	-0.4045	
В	1.6970	7.0462	-0.1419	
В	-0.6354	6.1331	1.2512	
В	0.6367	6.1331	-1.2513	
В	-0.6513	7.9029	1.2905	
В	-1.6956	7.0462	0.1418	
В	0.8926	7.0400	1.4453	
В	0.7915	8.4763	0.4009	
В	0.6527	7.9029	-1.2905	
В	-0.8912	7.0400	-1.4454	
В	-0.7902	8.4763	-0.4009	
Н	2.8651	6.8960	-0.2439	
Н	1.0502	5.3835	-2.0629	
Н	-1.5035	6.8956	-2.4463	
Н	1.1178	8.4771	-2.2167	
Н	-1.3649	9.4705	-0.6927	
Н	1.3663	9.4705	0.6927	
Н	1.5049	6.8955	2.4462	
Н	-1.0489	5.3836	2.0629	
Н	-2.8637	6.8961	0.2438	Η
Н	-1.1164	8.4771	2.2166	С
С	-1.4277	4.4078	-0.7504	С
С	-1.4214	3.7867	-1.9797	S
S	-2.3717	3.4866	0.4010	С
С	-2.1707	2.5886	-2.0121	Η
Н	-0.9071	4.1957	-2.8400	С
С	-2.7617	2.2775	-0.8022	Η
Н	-2.2917	1.9829	-2.9027	С
С	1.4290	4.4077	0.7504	С
С	1.4228	3.7869	1.9798	S
S	2.3725	3.4862	-0.4011	С
С	2.1719	2.5885	2.0122	Η
Н	0.9088	4.1960	2.8402	С
С	2.7626	2.2772	0.8022	Η
Н	2.2930	1.9830	2.9028	С
С	3.5907	1.1409	0.4679	С
С	4.4166	0.9597	-0.6264	Η
S	3.6405	-0.2664	1.5113	С
С	5.0917	-0.2792	-0.6277	С
Н	4.5497	1.7127	-1.3946	Η
С	4.7986	-1.0737	0.4667	С
Н	5.7917	-0.5760	-1.4005	Ν
С	-3.5902	1.1415	-0.4680	С
С	-4.4164	0.9607	0.6261	Ν
S	-3.6401	-0.2660	-1.5112	С
С	-5.0918	-0.2781	0.6275	Ν
Н	-4.5494	1.7138	1.3942	С
С	-4.7987	-1.0728	-0.4667	Ν



-5.7920	-0.5746	1.4002
-5.3068	-2.3798	-0.7950
-5.1206	-3.1093	-1.9638
-6.2954	-3.2635	0.3461
-5.7663	-4.3535	-1.9460
-4.5443	-2.7420	-2.8047
-6.4637	-4.6143	-0.7689
-5.7412	-5.0599	-2.7684
5.3064	-2.3808	0.7951
5.1203	-3.1101	1.9640
6.2945	-3.2649	-0.3461
5.7656	-4.3544	1.9462
4.5443	-2.7424	2.8050
6.4627	-4.6156	0.7691
5.7405	-5.0607	2.7688
7.1942	-5.8139	0.5341
7.9260	-6.2144	-0.5571
7.1614	-6.5208	1.3598
-7.1956	-5.8124	-0.5340
-7.9277	-6.2125	0.5571
-7.1629	-6.5193	-1.3596
8.5773	-7.4881	-0.5451
9.1039	-8.5262	-0.5273
8.0846	-5.4237	-1.7361
8.2120	-4.7776	-2.6969
-8.0864	-5.4216	1.7360
-8.2137	-4.7754	2.6968
-8.5794	-7.4860	0.5452
-9.1063	-8.5240	0.5274



Figure S7. TD-DFT calculation results for 1b and 2b. Energy levels and isodensity plots for HOMO-LUMOs are shown.



Figure S8. TD-DFT calculation results for 1c and 2c. Energy levels and isodensity plots for HOMO-LUMOs are shown.



**Figure S9.** Diagram of calculated LUMO energy levels vs dihedral angles of C–C bonds of the carborane cage with C–C bonds in the thiophene rings.

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