

## GREENER APPROACH OF SYNTHESIS OF AZO DERIVATIVES AND BIS- PYRAZOLE DERIVATIVES ALONG WITH ANTIMICROBIAL SCREENING

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### ABSTRACT

A clean, efficient and solvent free method for the synthesis of Azo compounds was carried out with the help of PbO nanoparticles. They have been synthesized by the reaction between 1-(N-methylpyridin-2-yl)pyrrolidine/ piperidine -2,5/6-dione and propionaldehyde at room temperature with the help of green chemistry. PbO nanoparticles employed as an efficient catalytic tool due to efficient, renewable and eco-friendly heterogeneous characteristics. The bis pyrazoles are synthesized by a simple eco-friendly microwave instigated solvent free synthesis. It was carried out by the reaction of corresponding bis-chalcones : 3,4/5-bis((E)-2-chlorobenzylidene)-1-(N-methylpyridin-2-yl)pyrrolidine/ piperidine -2,5/6-dione with hydrazine hydrate in presence of neutral Alumina. This method furnished various advantages, such as straight forward work-up procedure,

environmentally benevolent, neutral condition and high yield. All the derivatives were characterized and interpreted for antimicrobial activities.

**KEYWORDS:** Green Chemistry, clean, Solvent free, pyrrolidine-2,5-dione, piperidine-2,6-dione, propionaldehyde, hydrazine hydrate.

### INTRODUCTION

It is recognized that multi-functionalized benzopyrans and their derivatives are a very important class compounds. They have been broadly applicable as medicine intermediates

because their synergetic pharmacological and biological properties, such as anticoagulant, antibacterial, anticancer, spasmolytic, diuretic, hypnotic and insecticide.<sup>[1-7]</sup> Generally, substituted pyrazole and their derivatives are synthesized in organic solvents<sup>[8-10]</sup> which can be used as important pharmaceuticals and agricultural chemicals. Additionally, multi-substitution benzopyrans can contribute a structural unit for a group of natural products<sup>[11-13]</sup> and some of them along with their derivatives were signed up as photo-active materials.<sup>[14]</sup>

In recent years Pyrazoles are established as important class of heterocyclic community and have attractive achievements in both in organic synthesis and medicinal chemistry.<sup>[15-20]</sup> Pyrazoles are a very propitious synthetic intermediate which is source of an important pharmacophore exist in a variety of biologically active and potential therapeutic commixtures.<sup>[21-23]</sup> These admixtures have been mostly used in sententious antibacterial<sup>[24-28]</sup>, antipyretic<sup>[29-32]</sup>, antidepressant<sup>[33]</sup>, antiviral<sup>[34]</sup>, antitumor<sup>[35-38]</sup>, anti-inflammatory.<sup>[39-41]</sup> In addition various pyrazoles have been considered as the extracting and chelating reagents for different metal ions<sup>[42,43]</sup> and few of the pyrazolone derivatives are used in several marketable drugs for myocardial and brain ischemia.<sup>[44,45]</sup> Now a days research is focused on eco-friendly, solvent free efficient greener synthesis. In line of this few catalyst are invented for the synthesis of pyrazoles like heteropolyacids<sup>[46, 47]</sup>, acetic acid or piperidine<sup>[48]</sup>, Na<sup>+</sup>-MMT-[pmim]HSO<sub>4</sub><sup>[49]</sup>, cesium fluoride (CsF)<sup>[50]</sup>, 3-aminopropylated silica gel<sup>[51]</sup>, 1,3,5-tris(hydrogensulfato) benzene (THSB)<sup>[52]</sup>, sodium dodecyl sulfate (SDS)<sup>[53]</sup>, ZnAl<sub>2</sub>O<sub>4</sub> nanoparticles<sup>[54]</sup>, silica-bonded S-sulfonic acid (SBSSA)<sup>[55]</sup>, LiOH·H<sub>2</sub>O<sup>[56]</sup>, [Cu(3,4-tmtppa)](MeSO<sub>4</sub>)<sub>4</sub><sup>[57]</sup> ceric ammonium nitrate (CAN), benzyltriethylammonium chloride<sup>[58]</sup> and so on. However, it is well-known that several of the methods reported above faces challenges like strong basic or acidic conditions, difficulty in handling and separation of catalyst, use of costly catalysts long reaction time and huge solvent consumption. These transactions include possibility of side reactions which restrict their usage in practical applications. Wherefore, it is very important to ponder and develop novel eco-friendly efficient and scalable synthetic routes which are able to construct pyrazolone derivatives in high yields.

The current trends of research are focused on the applications of the solid support Al<sub>2</sub>O<sub>3</sub> and silica gel SiO<sub>2</sub>. They have pulled out much attention as not only they could be used as a catalyst for many organic transactions which providing high yields but also cheaper and non-

toxic. Also, It is found that  $\text{Al}_2\text{O}_3$  and  $\text{SiO}_2$  have greatly simplified the workup in a solvent-free procedures.<sup>[59,60]</sup>

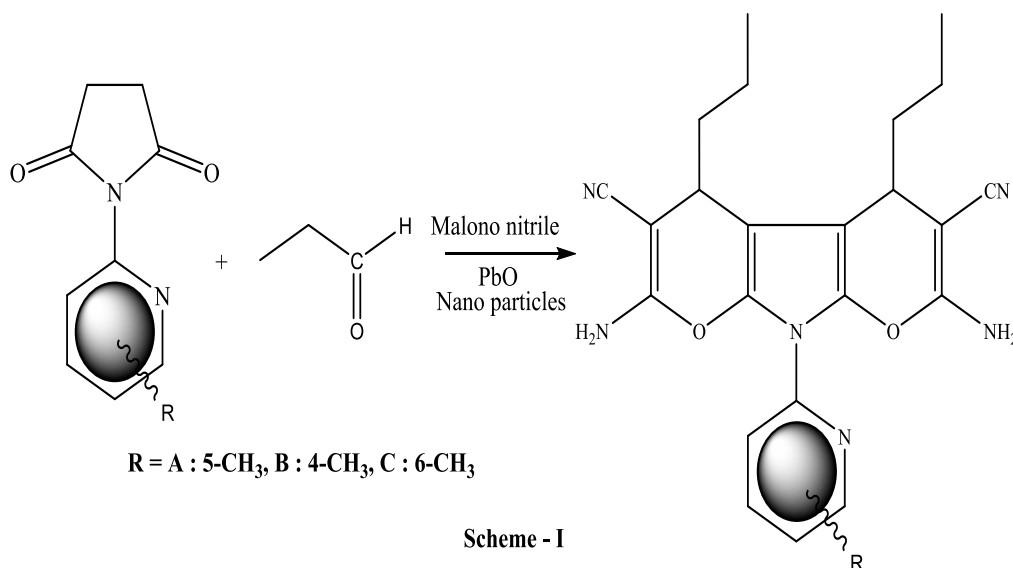
## EXPERIMENTAL

### MATERIAL METHODS

Melting points were recorded in open glass capillaries and were uncorrected. The chemical structures of the obtained compounds were confirmed by spectral analyses. IR spectra in KBr pellets were obtained on Simadzu and ATR Bruker alpha FT-IR spectrophotometer.  $^1\text{H}$  NMR spectra were obtained on and 500.13 MHz by Bruker spectrophotometer. The chemical shifts were reported as parts per million (ppm) with  $(\text{CH}_3)_4\text{Si}$  (TMS) as an internal standard. Signal multiplicities are represented by: s (singlet), d (doublet), t (triplet), m (multiplet). The purity of compound was checked by thin layer chromatography which was performed by using pre-coated silica gel aluminium plates with mixture of diethyl ether and ethyl acetate 7:3 proportion. Anti-microbial and Anti-fungal activities were carried out by Agar diffusion assay (Disk diffusion method, Disk size 6 mm). PbO nanoparticle were synthesized by taking a mixture of 10 ml of 0.1N sodium hydroxide and 0.025 mole citric acid in distilled water was added to methanolic solution of 0.02 mole lead nitrate. The reaction mixture was continuously stirred with the help of magnetic stirrer for 2 hours at room temperature. The white polycrystalline product was obtained which is filtered, washed with distilled water and dried at  $110^\circ\text{C}$  for 2 hours. The dried solid product was calcinized at  $500^\circ\text{C}$  for 2 hours. During this process, the PbO nanoparticle which has white colour earlier turned to pale yellow colour. All the compounds (7a-f and 10a-f) were synthesized from the corresponding Succinic and Glutaric Anhydride derivatives and commercially purchased propionaldehyde, neutral alumina ( $\text{Al}_2\text{O}_3$ ) and ethanol.

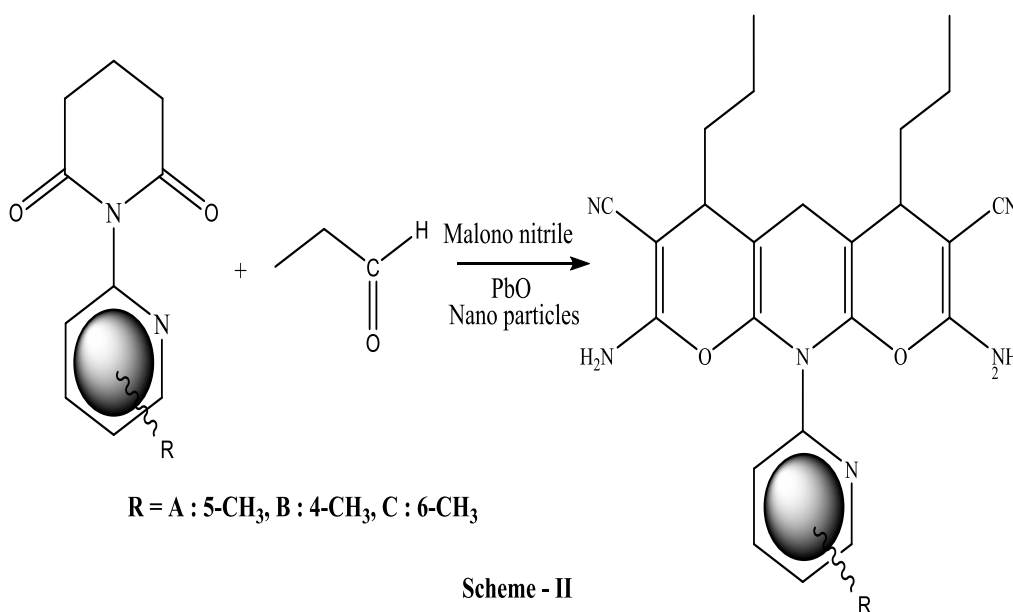
### General Procedure of synthesis

Preparation of Azo derivatives: 2,7-diamino-9-(N-methylpyridin-2-yl)-4,5-dipropyl-5,9-dihydro-4H-dipyran[2,3-b:3',2'-d]pyrrole-3,6-dicarbonitrile (7a-c): A mixture of 0.01 mole N-phenyl pyrrolidine-2,5-dione, 0.02 mole propionaldehyde, 0.02 mole malononitrile and 100 mg PbO nanoparticles were ground at a room temperature with a mortar and pestle. The reaction was monitored by thin layer chromatography (TLC). After completion of reaction, the product was washed with distilled water. The novel developed compounds were dried and recrystallized from ethanol to afford pure compounds with high yield (**Scheme – I**).



**Preparation of Azo (Dipyrano): 2,8-diamino-10-(N-methylpyridin-2-yl)-4,6-dipropyl-6,10-dihydro-4H,5H-dipyrano[2,3-b:3',2'-e]pyridine-3,7-dicarbonitrile (7d-f)**

A mixture of 0.01 mole N-phenyl piperidine-2,6-dione, 0.02 mole propionaldehyde, 0.02 mole malononitrile and 100 mg PbO nanoparticles were ground at a room temperature with a mortar and pestle. The reaction was monitored by thin layer chromatography (TLC). After completion of reaction, the product was washed with distilled water. The novel developed compounds were dried and recrystallized from ethanol to afford pure compounds with high yield (Scheme – II).



**Physicochemical and analytical data for compounds 7a-f*****2,7-diamino-9-(5-methylpyridin-2-yl)-4,5-dipropyl-5,9-dihydro-4H-dipyrano[2,3-b:3',2'-d]pyrrole-3,6-dicarbonitrile (7a)***

Yellow Orange Solid, Yield (82.14%), M. P. 268-70°C, M.F. C<sub>24</sub>H<sub>26</sub>O<sub>2</sub>N<sub>6</sub> M.W.430.50, Composition: C (66.29%) H (6.35%) N (19.28%); IR (KBr): C≡N:2137.50; -N-H: 3334.46; aromatic ring: 683.60; -CH<sub>3</sub>: 3214.75; C-N (Aliphatic): 1154.36; C-N (Aromatic): 1222.63; -C-C- Stretch in a ring(2-Peaks): 1637.43, 1539.22; -CH<sub>3</sub> bend: 1492.79, 1384.71 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 0.90 (d, 6H), 1.22 (t, 4H, Methylene), 1.51 (t,4H, Methylene), 2.12 (s,2H, methine),6.86 (s, 4H,-NH<sub>2</sub>), 7.30-8.49 (m, 3H, aromatic), 2.37 (s, 3H, CH<sub>3</sub>-pyridine).

***2,7-diamino-9-(4-methylpyridin-2-yl)-4,5-dipropyl-5,9-dihydro-4H-dipyrano[2,3-b:3',2'-d]pyrrole-3,6-dicarbonitrile (7b)***

Yellow Orange Solid, Yield (78.15%), M. P. 312-14°C, M.F. C<sub>24</sub>H<sub>26</sub>O<sub>2</sub>N<sub>6</sub> M.W.430.50, Composition: C (66.39%) H (6.26%) N (19.41%); IR (KBr): C≡N:2141.96; -N-H: 3330.28; aromatic ring: 683.41; -CH<sub>3</sub>: 2921.49; C-N (Aliphatic): 1048.07; C-N (Aromatic): 1222.07; -C-C- Stretch in a ring(2-Peaks): 1620.98, 1539.73; -CH<sub>3</sub> bend: 1492.99, 1383.42 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 0.94 (d, 6H), 1.24 (t,4H, Methylene), 1.52 (t,4H, Methylene), 2.12 (s,2H, methine),6.80 (s, 4H,-NH<sub>2</sub>), 7.34-8.54 (m, 3H, aromatic), 2.37 (s, 3H, CH<sub>3</sub>-pyridine).

***2,7-diamino-9-(6-methylpyridin-2-yl)-4,5-dipropyl-5,9-dihydro-4H-dipyrano[2,3-b:3',2'-d]pyrrole-3,6-dicarbonitrile (7c)***

Yellow Orange Solid, Yield (78.72%), M. P. 188-90°C, M.F. C<sub>24</sub>H<sub>26</sub>O<sub>2</sub>N<sub>6</sub> M.W.430.50, Composition: C (66.62%) H (6.19%) N (19.37%); IR (KBr): C≡N: 2164.50; -N-H: 3337.62; aromatic ring: 687.72; -CH<sub>3</sub>: 2968.39; C-N (Aliphatic): 1058.76; C-N (Aromatic): 1228.01 -C-C- Stretch in a ring(2-Peaks): 1646.36, 1495.52; -CH<sub>3</sub> bend: 1457.24, 1384.24 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 0.90 (d, 6H), 1.27 (t,4H, Methylene), 1.54 (t,4H, Methylene), 2.12 (s,2H, methine),6.91 (s, 4H,-NH<sub>2</sub>), 7.04-7.79 (m, 3H, aromatic), 2.64 (s, 3H, CH<sub>3</sub>-pyridine).

***2,8-diamino-10-(5-methylpyridin-2-yl)-4,6-dipropyl-6,10-dihydro-4H,5H-dipyrano[2,3-b:3',2'-e]pyridine-3,7-dicarbonitrile (7d)***

Yellow Orange Color Solid, Yield (78.49%), M. P. 184-86°C, M.F. C<sub>25</sub>H<sub>28</sub>O<sub>2</sub>N<sub>6</sub> M.W.444.52, Composition: C (67.02%) H (6.89%) N (18.26%); IR (KBr): C≡N:2135.17; -N-H: 3332.09;

aromatic ring: 684.53; -CH<sub>2</sub>: 3001.18; -CH<sub>3</sub>: 2927.33; C-N (Aliphatic): 1153.27; C-N (Aromatic): 1298.31; -C-C- Stretch in a ring(2-Peaks): 1730.05, 1532.34; -CH<sub>3</sub> bend: 1451.66; -CH<sub>2</sub> bend: 1380.28 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 0.89 (d, 6H), 1.17 (t,4H, Methylene), 2.44 (t,4H, Methylene), 2.56 (s,2H, methine), 3.53 (s, 2H, -CH<sub>2</sub>), 6.56 (s, 4H,-NH<sub>2</sub>), 7.19-7.75 (m, 3H, aromatic), 2.51 (s, 3H, CH<sub>3</sub>-pyridine).

***2,8-diamino-10-(4-methylpyridin-2-yl)-4,6-dipropyl-6,10-dihydro-4H,5H-dipyrano[2,3-b:3',2'-e]pyridine-3,7-dicarbonitrile (7e)***

Yellow Orange Color Solid, Yield (72.41%), M. P. 166-68°C, M.F. C<sub>25</sub>H<sub>28</sub>O<sub>2</sub>N<sub>6</sub> M.W.444.52, Composition: C (67.19%) H (6.93%) N (18.06%); IR (KBr): C≡N:2140.5; -N-H: 3331.19; aromatic ring: 685.88; -CH<sub>2</sub>: 3003.54; -CH<sub>3</sub>: 2928.51; C-N (Aliphatic): 1050.34; C-N (Aromatic): 1290.18; -C-C- Stretch in a ring(2-Peaks): 1740.22, 1531.94; -CH<sub>3</sub> bend: 1453.53; -CH<sub>2</sub> bend: 1379.36 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 0.91 (d, 6H), 1.20 (t,4H, Methylene), 2.39 (t,4H, Methylene), 2.35 (s,2H, methine), 3.53 (s, 2H, -CH<sub>2</sub>), 6.60 (s, 4H,-NH<sub>2</sub>), 7.22-7.35 (m, 3H, aromatic), 2.33 (s, 3H, CH<sub>3</sub>-pyridine).

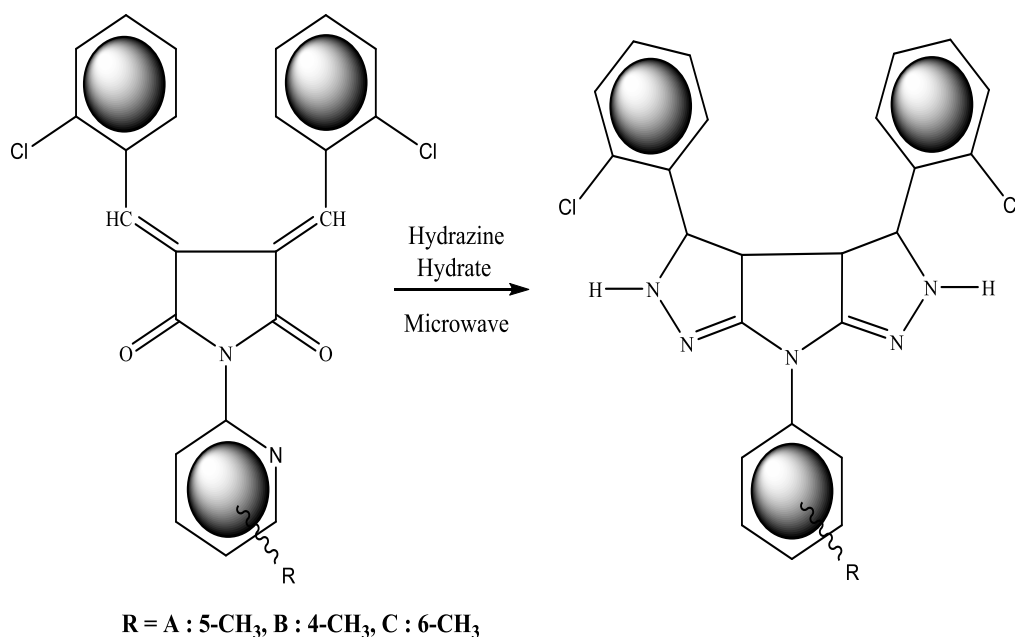
***2,8-diamino-10-(6-methylpyridin-2-yl)-4,6-dipropyl-6,10-dihydro-4H,5H-dipyrano[2,3-b:3',2'-e]pyridine-3,7-dicarbonitrile (7f)***

Pastel Orange Solid, Yield (78.22%), M. P. 240-42°C, M.F. C<sub>25</sub>H<sub>28</sub>O<sub>2</sub>N<sub>6</sub> M.W.444.52, Composition: C (66.98%) H (6.11%) N (18.19%); IR (KBr): C≡N:2137.05; -N-H: 3327.75; aromatic ring: 683.34; -CH<sub>2</sub>: 3002.44; -CH<sub>3</sub>: 2925.64; C-N (Aliphatic): 1156.90; C-N (Aromatic): 1300.21; -C-C- Stretch in a ring(2-Peaks): 1744.42, 1496.28; -CH<sub>3</sub> bend: 1452.13; -CH<sub>2</sub> bend: 1388.53 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 0.94 (d, 6H), 1.15 (t,4H, Methylene), 2.40 (t,4H, Methylene), 2.64 (s,2H, methine), 3.59 (s, 2H, -CH<sub>2</sub>), 6.69 (s, 4H,-NH<sub>2</sub>), 7.25-7.64 (m, 3H, aromatic), 2.45 (s, 3H, CH<sub>3</sub>-pyridine).

### General Procedure of Synthesis

**Preparation of Pyrazole: 3,4-bis(2-chlorophenyl)-7-(N-methylpyridin-2-yl)-3,3a,3b,4,5,7-hexahydro-2H-pyrrolo[2,3-c:5,4-c']dipyrzole (13a-c)**

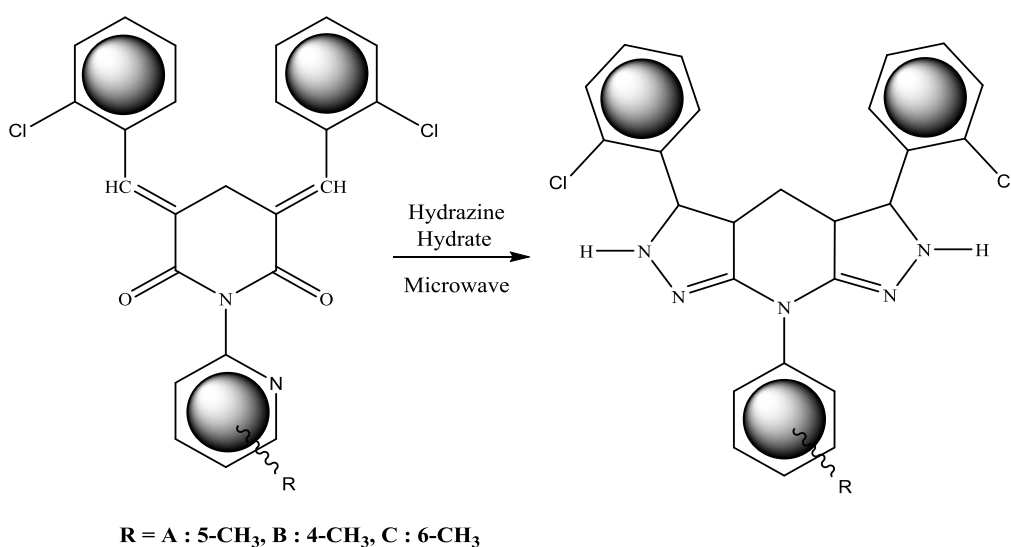
The bis-pyrazole (13a-c) derivatives were synthesized by the mixture of 0.01 mole of N-phenyl pyrrolidine-2, 5-dione and 0.02 mole of aromatic aldehyde in 1 gm of neutral Al<sub>2</sub>O<sub>3</sub> with the help of microwave irradiations. This mixture is maintained in microwave at 800W power for 4-6 minutes in solvent free condition. The novel developed compounds were recrystallized from ethanol (Scheme – III).



Scheme - III

**Preparation of Pyrazole: 3,5-bis(2-chlorophenyl)-8-(N-methylpyridin-2-yl)-2,3,3a,4,4a,5,6,8-octahydrodipyrzolo[3,4-b:4',3'-e]pyridine (13d-f)**

The bis-pyrazole (13d-f) derivatives were synthesized by the mixture of 0.01 moles of N-phenyl piperidine-2,6-dione and 0.02 mole of aromatic aldehyde in 1 gm of neutral Al<sub>2</sub>O<sub>3</sub> with the help of microwave irradiations. This mixture is maintained in microwave at 800W power for 5-8 minutes in solvent free condition. The novel developed compounds were recrystallized from ethanol (Scheme – IV).



Scheme - IV

**Physicochemical and analytical data for compounds 13a-f*****3,4-bis(2-chlorophenyl)-7-(5-methylpyridin-2-yl)-3,3a,3b,4,5,7-hexahydro-2H-pyrrolo[2,3-c:5,4-c']dipyrazole (13a)***

Green Brown Solid, Yield (71.33%), M. P. 136-38°C, M.F. C<sub>24</sub>H<sub>20</sub>N<sub>6</sub>Cl<sub>2</sub> M.W.463.36, Composition: C (63.25%) H (3.81%) N (14.47%); IR (KBr): -C-Cl:645.06; -N-H: 3177.50; >C=N: 1655.70; aromatic ring (2-Peaks): 3063.07, 824.28; -CH<sub>3</sub>: 3004.38; C-N (Aliphatic): 1209.99; C-N (Aromatic): 1274.57; -C-C- Stretch in a ring(2-Peaks): 1564.02, 1506.82; -CH<sub>3</sub> bend: 1467.09, 1371.66 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 2.20 (t, 2H, Methine), 3.40 (d, 2H, Methine), 9.85 (s, 2H,-N-H), 6.64-7.88 (m, 11H, aromatic), 2.28 (s, 3H, CH<sub>3</sub>-pyridine).

***3,4-bis(2-chlorophenyl)-7-(4-methylpyridin-2-yl)-3,3a,3b,4,5,7-hexahydro-2H-pyrrolo[2,3-c:5,4-c']dipyrazole (13b)***

Faint Clay Brown Solid, Yield (76.24%), M. P. 194-96°C, M.F. C<sub>24</sub>H<sub>20</sub>N<sub>6</sub>Cl<sub>2</sub> M.W.463.36, Composition: C (63.30%) H (3.37%) N (14.55%); IR (KBr): -C-Cl:644.71; -N-H: 3150.16; >C=N: 1655.17; aromatic ring (2-Peaks): 3051.27, 825.15; -CH<sub>3</sub>: 3003.82; C-N (Aliphatic): 1211.31; C-N (Aromatic): 1271.43; -C-C- Stretch in a ring(2-Peaks): 1566.17, 1528.29; -CH<sub>3</sub> bend: 1465.26, 1373.61 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 2.28 (t, 2H, Methine), 3.47 (d, 2H, Methine), 9.96 (s, 2H,-N-H), 6.57-7.95 (m, 11H, aromatic), 2.29 (s, 3H, CH<sub>3</sub>-pyridine).

***3,4-bis(2-chlorophenyl)-7-(6-methylpyridin-2-yl)-3,3a,3b,4,5,7-hexahydro-2H-pyrrolo[2,3-c:5,4-c']dipyrazole (13c)***

Saffron Yellow Solid, Yield (63.20%), M. P. 326-28°C, M.F. C<sub>24</sub>H<sub>20</sub>N<sub>6</sub>Cl<sub>2</sub> M.W.463.36, Composition: C (63.61%) H (3.76%) N (14.39%); IR (KBr): -C-Cl:643.26; -N-H: 3165.52; >C=N: 1664.68; aromatic ring (2-Peaks): 3048.18, 823.25; -CH<sub>3</sub>: 3001.55; C-N (Aliphatic): 1212.44; C-N (Aromatic): 1275.36; -C-C- Stretch in a ring(2-Peaks): 1532.77, 1572.25; -CH<sub>3</sub> bend: 1461.12, 1372.45 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 2.17 (t, 2H, Methine), 3.08 (d, 2H, Methine), 9.95 (s, 2H,-N-H), 6.27-7.45 (m, 11H, aromatic), 2.36 (s, 3H, CH<sub>3</sub>-pyridine).

***3,5-bis(2-chlorophenyl)-8-(5-methylpyridin-2-yl)-2,3,3a,4,4a,5,6,8-octahydrodipyrazolo[3,4-b:4',3'-e]pyridine (13d)***

Honey Yellow Solid, Yield (77.38%), M. P. 332-34°C, M.F. C<sub>25</sub>H<sub>22</sub>N<sub>6</sub>Cl<sub>2</sub> M.W.477.38, Composition: C (63.05%) H (4.91%) N (17.21%); IR (KBr): -C-Cl:643.64; -N-H: 3176.69;



>C=N: 1654.66; aromatic ring (2-Peaks): 3002.00, 856.72; -CH<sub>3</sub>: 2856.80; -CH<sub>2</sub>: 2827.62; C-N (Aliphatic): 1210.84; C-N (Aromatic): 1275.79; -C-C- Stretch in a ring(2-Peaks): 1594.73, 1563.93; -CH<sub>3</sub> bend: 1436.03, 1365.69; -CH<sub>2</sub> bend: 1466.81 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 2.26 (m, 2H, methine), 3.79 (m, 2H, methine), 1.69 (m, 2H, methylene), 9.94 (s, 2H,-N-H), 6.51-7.83 (m, 11H, aromatic), 2.14 (s, 3H, CH<sub>3</sub>-pyridine).

***3,5-bis(2-chlorophenyl)-8-(4-methylpyridin-2-yl)-2,3,3a,4,4a,5,6,8-octahydrodipyrazolo[3,4-b:4',3'-e]pyridine (13e)***

Clay Brown Solid, Yield (65.74%), M. P. 278-80°C, M.F. C<sub>25</sub>H<sub>22</sub>N<sub>6</sub>Cl<sub>2</sub> M.W.477.38, Composition: C (62.09%) H (4.21%) N (17.18%); IR (KBr): -C-Cl:641.28; -N-H: 3210.31; >C=N: 1655.41; aromatic ring (2-Peaks): 3001.98, 855.33; -CH<sub>3</sub>: 2879.28; -CH<sub>2</sub>: 2825.19; C-N (Aliphatic): 1209.75; C-N (Aromatic): 1274.39; -C-C- Stretch in a ring(2-Peaks): 1580.29, 1565.74; -CH<sub>3</sub> bend: 1434.28, 1365.47; -CH<sub>2</sub> bend: 1463.08 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 2.21 (m, 2H, methine), 3.77 (m, 2H, methine), 1.75 (m, 2H, methylene), 9.93 (s, 2H,-N-H), 6.49-7.86 (m, 11H, aromatic), 2.16 (s, 3H, CH<sub>3</sub>-pyridine).

***3,5-bis(2-chlorophenyl)-8-(6-methylpyridin-2-yl)-2,3,3a,4,4a,5,6,8-octahydrodipyrazolo[3,4-b:4',3'-e]pyridine (13f)***

Deep Orange Solid, Yield (75.61%), M. P. 334-36°C, M.F. C<sub>25</sub>H<sub>22</sub>N<sub>6</sub>Cl<sub>2</sub> M.W.477.38, Composition: C (62.18%) H (4.22%) N (17.01%); IR (KBr): -C-Cl:640.65; -N-H: 3202.88; >C=N: 1657.50; aromatic ring (2-Peaks): 3004.91, 858.21; -CH<sub>3</sub>: 2949.15; -CH<sub>2</sub>: 2926.74; C-N (Aliphatic): 1208.87; C-N (Aromatic): 1273.11; -C-C- Stretch in a ring(2-Peaks): 1587.44, 1568.37; -CH<sub>3</sub> bend: 1435.39, 1373.13; -CH<sub>2</sub> bend: 1463.94 cm<sup>-1</sup>. <sup>1</sup>H NMR (500.13 MHz, DMSO, δ ppm): 2.37 (m, 2H, methine), 3.86 (m, 2H, methine), 1.71 (m, 2H, methylene), 9.91 (s, 2H,-N-H), 6.32-7.62 (m, 11H, aromatic), 2.21 (s, 3H, CH<sub>3</sub>-pyridine).

## RESULTS AND DISCUSSION

### Chemistry

The series of Azo derivatives 7a-f were synthesized by the reaction of 1-(N-methylpyridin-2-yl)pyrrolidine/ piperidine -2,5/6-dione, propionaldehyde and malononitrile with the help of PbO nanoparticles. The formation of Azo derivatives was confirmed by IR, <sup>13</sup>C NMR and <sup>1</sup>H NMR and elemental analysis. The series of bis-pyrazole derivatives 13a-f were synthesized by the reaction of bis chalcones 3,4/5-bis((E)-2-chlorobenzylidene)-1-(N-methylpyridin-2-yl)pyrrolidine/ piperidine -2,5/6-dione and hydrazine hydrate in presence of neutral Al<sub>2</sub>O<sub>3</sub>

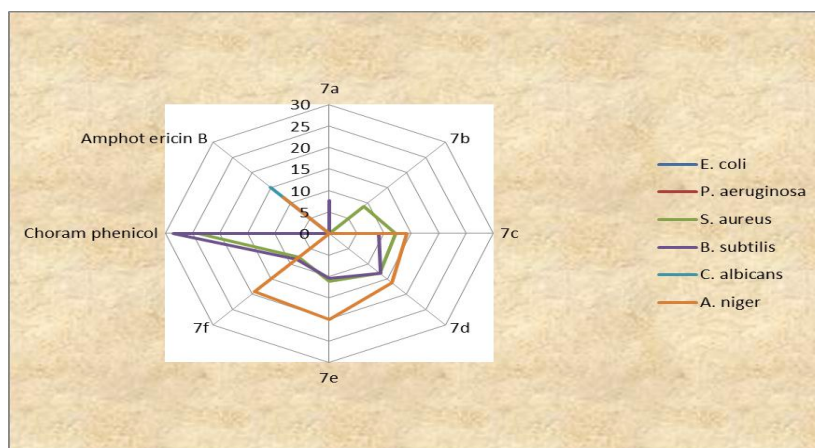
with the help of microwave irradiations. The formation of bis-pyrazoles was confirmed by IR,  $^{13}\text{C}$  NMR and  $^1\text{H}$  NMR and elemental analysis.

### Antimicrobial Activities

All the synthesized Azo derivatives 7a-f and bis-pyrazole derivatives 13a-f were screened for their antibacterial activity against gram positive bacteria *Staphylococcus aureus* (NCIM 2079), *Bacillus subtilis* (NCIM 2250) and gram negative bacteria *Escherichia coli* (NCIM 2109), *Pseudomonas aeruginosa* (NCIM 2036) using DMSO solvent. All these novel synthesized compounds were screened against Fungi (Yeast) *Candida albicans* (NCIM 3471) and *Aspergillus niger* (NCIM 545). The bacterial cultures were purchased from NCIM: National Collection of Industrial Microorganisms, National Chemical Laboratory (NCL), Pune 411008 [India]. Some of the compound showed moderate to good activities against gram positive bacteria *S. aureus* and synergetic activities against Fungi *A. niger* as shown in the Table –I, II and Graph –I, II.

**Table-I: Antimicrobial activities of Azo derivatives 7a-f.**

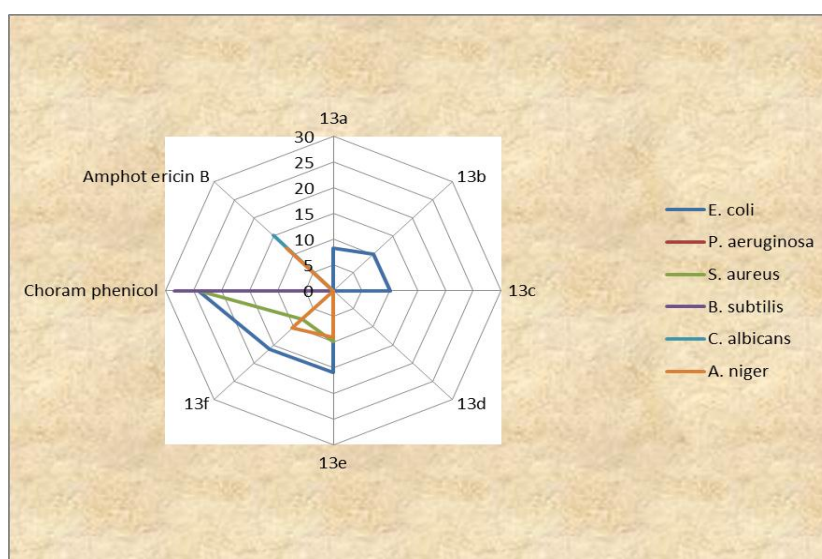
Sr. No.	Sample	<i>E. coli</i>	<i>P. aeruginosa</i>	<i>S. aureus</i>	<i>B. subtilis</i>	<i>C. albicans</i>	<i>A. niger</i>
		Mean±SD	Mean ±SD	Mean ±SD	Mean ±SD	Mean ±SD	Mean ±SD
1	7a	- - -	- - -		7.66±0.20	- - -	- - -
2	7b	- - -	- - -	8.96±0.09	- - -	- - -	- - -
3	7c	- - -	- - -	12.06±0.07	9.05±0.13	- - -	14.2±0.07
4	7d	- - -	- - -	13.09±0.02	13.19±0.06	- - -	16.2±0.01
5	7e	- - -	- - -	11.09±0.14	10.49±0.05	- - -	20±0.12
6	7f	- - -	- - -	7.86±0.11	8.40±0.06	- - -	19.2±0.11
	<b>Choram phenicol</b>	<b>24.09±0.10</b>	<b>14.39±0.07</b>	<b>23.92±0.17</b>	<b>28.43±0.29</b>	NA	NA
	<b>Amphot ericin B</b>	NA	NA	NA	NA	<b>15.21±0.15</b>	<b>11.8±0.08</b>



**Graph-I: Antimicrobial activities of Azo derivatives 7a-f.**

Table-II: Antimicrobial activities of bis pyrazole derivatives 13a-f.

Sr. No.	Sample	<i>E. coli</i>	<i>P. aeruginosa</i>	<i>S. aureus</i>	<i>B. subtilis</i>	<i>C. albicans</i>	<i>A. niger</i>
		Mean±SD	Mean ±SD	Mean ±SD	Mean ±SD	Mean ±SD	Mean ±SD
1	13a	8.28±0.09	- - -	- - -	- - -	- - -	- - -
2	13b	10.01±0.18	- - -	- - -	- - -	- - -	- - -
3	13c	10.08±0.13	- - -	- - -	- - -	- - -	- - -
4	13d	- - -	- - -	- - -	- - -	- - -	- - -
5	13e	15.93±0.14	- - -	9.85±0.15	- - -	- - -	9.11±0.04
6	13f	16.1±0.18	- - -	7.82±0.03	- - -	- - -	10.3±0.16
	<b>Choram phenicol</b>	<b>24.09±0.10</b>	<b>14.39±0.07</b>	<b>23.92±0.17</b>	<b>28.43±0.29</b>	NA	NA
	<b>Amphot ericin B</b>	NA	NA	NA	NA	<b>15.21±0.15</b>	<b>11.8±0.08</b>



Graph-II: Antimicrobial activities of bis pyrazole derivatives 13a-f.

## CONCLUSION

The greener path used catalyst PbO nanoparticle which is solvent-free and produced sensible yield. 13e and 13f have showed ameliorate activities against *S. aureus*. Also, the series of compounds 13a-c reveals excellent and congenial activities against *A. niger* strain.

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