

PROMISING CASTING ALUMINUM ALLOYS WITHOUT REQUIREMENT FOR HEAT TREATMENT

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Abstract

Three kinds of aluminum alloys based on Al-Zn-Mg system and doped with Ca, Ni and Ce were studied. Determination of hot tearing tendency, phase composition and structure formation investigation were conducted. Alloying with the elements of Ca, Ni, Ce group is effective to increase casting properties that is better than those of a commercial A206 alloy. The structure after slow cooling during sand mold casting contains mainly needle like Al₃Fe. But after permanent mold casting the iron containing phases are only ternary phases Al₁₀CaFe₂, Al₉FeNi and Al₁₀CeFe₂. Most of Zn and Mg content are in aluminum solid solution as a result of casting that cause of appropriate mechanical properties under as-cast state. Moreover the high content of iron can allow us to include new alloys into the recycled type which means the opportunity to be produced using low grade aluminum or can scrap.

Keywords: casting, Al-Zn-Mg system, nickel, calcium, cerium

1. INTRODUCTION

Most of casting aluminum alloys belong to Al-Si (4xx) system [1, 2]. They consist of great many eutectic constituents which make them well-castable by any methods into almost any mold with any configuration. Also these alloys are well known as recyclable that means that they are able to be produced with use a secondary stock [2, 3]. Thus, they usually contain iron. The non-heat-treatable Al-Si alloys are usually subjected to high-pressure die casting (HPDC) and their ultimate tensile strength (UTS) under as-cast state does not exceed 200 MPa or they have low ductility [2, 4, 5]. Some kinds of Al-Mg (5xx) alloys developed in [6-8] have a very high ductility after HPDC under as-cast state provided by limitation of iron content. In spite of high strain rate these alloys are not secondary type and also they have low yield strength (YS) of less than 200 MPa that is common for 5xx alloys [2].

Al-Zn-Mg (7xxx) alloys have not been considered as non-heat-treatable casting alloys yet. The matrix system provides high mechanical properties after ageing but due to lack of eutectic liquid they usually produced as wrought products [1]. Also a strict limitation of iron is necessary too [1, 9]. Nevertheless it looks promising because of some facts linked with their good weldability (in no-cooper containing alloys) and ability to obtain supersaturated solid solution after casting at ease [1]. Some works are devoted to both eutectic forming elements and iron addition [10, 11]. In one of these works [11] authors alloyed Al-Zn-Mg matrix with 0.41 % Fe and 0.55 % Ni to obtain Al₉FeNi ternary phase in structure of castings after crystallization under pressure. UTS value was about 500 MPa in T6 state. It could be guessed that Al-Zn-Mg-Ni-Fe alloys are the most studied among Al-Zn-Mg alloys with eutectics. Calcium is equally good to obtain eutectic. It is able to enhance corrosion resistance and make a density lower [12]. Al-Zn-Mg-Ca-Fe alloys were considered in an only study [10]. They showed a positive effect of iron on hardness due to its reaction with aluminum and calcium forming an Al₁₀CaFe₂ phase and (Al, Zn)₄Ca phase content reduction. No tensile properties were

presented. Cerium is also forming the same phases $Al_{10}CeFe_2$ and Al_4Ce of eutectic origin [14]. But it has never been presented in Al-Zn-Mg alloys. The only work considers Al-Zn-Mg-Cu alloy doped with no more than 0.4 % Ce [15]. Despite the positive grain refinement effect the coarse Al_8Cu_4Ce appears. Due to this fact and according to some recommendations [2] alloying with copper is not considered in this study.

This work aims to substantiate a principal opportunity to obtain new promising high strength aluminum alloys based on the Al-Zn-Mg-Ni(Ca, Ce)-Fe system for use under as-cast state by determination of casting properties, microstructure and mechanical properties investigation.

2. EXPERIMENT

An initial matrix system Al–5.5 % Zn–1.5 % Mg was alloyed with 0.5 % Fe and separately doped with 1 % Ni, 1 % Ca and 1 % Ce. These three kinds of alloys were obtained by smelting pure materials and master alloys: aluminum (purity of 99.8 %), zinc (purity of 100 %), magnesium (purity of 100 %), cerium (purity of 100 %), master alloys Al–10 % Fe, Al–15 % Ca and Al–20 % Ni. The melting temperature was no more than 850 °C and casting temperature was 720-730 °C. The melts purification was conducted by C_2Cl_6 powder injection covered with aluminum foil. The experimental chemical composition (**Table 1**) was determined by spectral analysis in ARL3460 emission spectrometer. It could be seen that the experimental concentrations are enough close to the theoretical ones.

Table 1 Chemical composition of experimental alloys

Alloy designation	Concentrations (wt. %)						
	Zn	Mg	Ni	Ca	Ce	Fe	Al
A1	5.61	1.65	0.84	-	-	0.44	Balance
A2	5.48	1.44	-	0.91	-	0.47	Balance
A3	5.60	1.38	-	-	1.11	0.45	Balance

Pencil-type hot tearing test castings with diameters of 16, 14, 12, and 10 mm were obtained. The cooling rate was at least 20 °C/s. The microstructure analysis was carried out by scanning electron microscopy (SEM) TESCAN VEGA 3 and electron microprobe analysis (Oxford AZtec). Specimens were polished using a diamond suspension with 9, 6, and 3 μ m fractions. They also were subjected to electrolytic etching at 12 V in 6 C_2H_5OH : 1 $HClO_4$: 1 glycerin water solution. To investigate equilibrium structure specimens after slow cooling (less than 5 °C/s) in sand mold were obtained. Also some bars permanent mold castings with a massive bob of about 600 g were obtained (**Figure 1**). They were mechanically processed for tensile test samples which were subjected to room temperature uniaxial test at Z250 Zwick/Roell machine in an as-cast state.

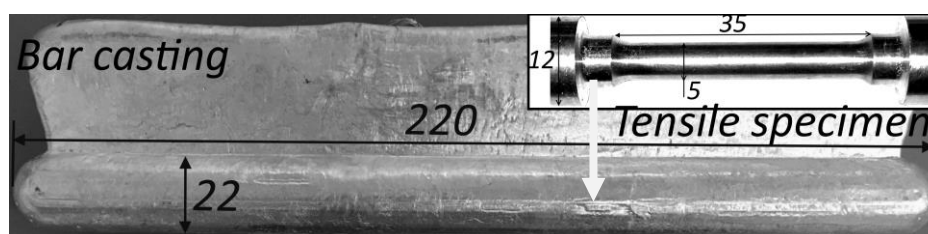


Figure 1 Bar permanent casting

3. RESULTS AND DISCUSSION

Hot tearing tendency appears due to low ductility in solid-liquid state. This state occurs during effective solidification range. This means a formation of solid body of casting that is too fragile and undergoes to shrinkage. A well-known temperature of non-equilibrium solidus in Al-Zn-Mg system is about 483 °C, therefore a solidification range is too wide compared to Al-Si alloys. At this temperature T phase ($\text{Al}_2\text{Mg}_3\text{Zn}_3$) appears. Due to dissolution of some Zn and Mg in (Al) there is a lack of eutectic liquid to fill the emerged cracks. In this case the alloying with iron and eutectic forming elements contributes to formation of non-soluble Al_9FeNi , $(\text{Al,Zn})_4\text{Ca}$, $\text{Al}_{10}\text{CaFe}_2$, $\text{Al}_{10}\text{CeFe}_2$ and Al_4Ce phases of equilibrium eutectic origin at temperature above 500 °C [2,10,14]. At implemented solidification rates a lower border of the effective solidification range may shift to equilibrium solidus border. Thus, a brittleness temperature range is narrower and crystallizing equilibrium eutectic which crystallizes the last is relatively enough to fill hot cracks. This was confirmed during hot tearing tendency test. **Figure 2** indicates on better casting properties of the experimental alloys A1, A2, A3 than those of a reference commercial A206 (Al-5 % Cu).

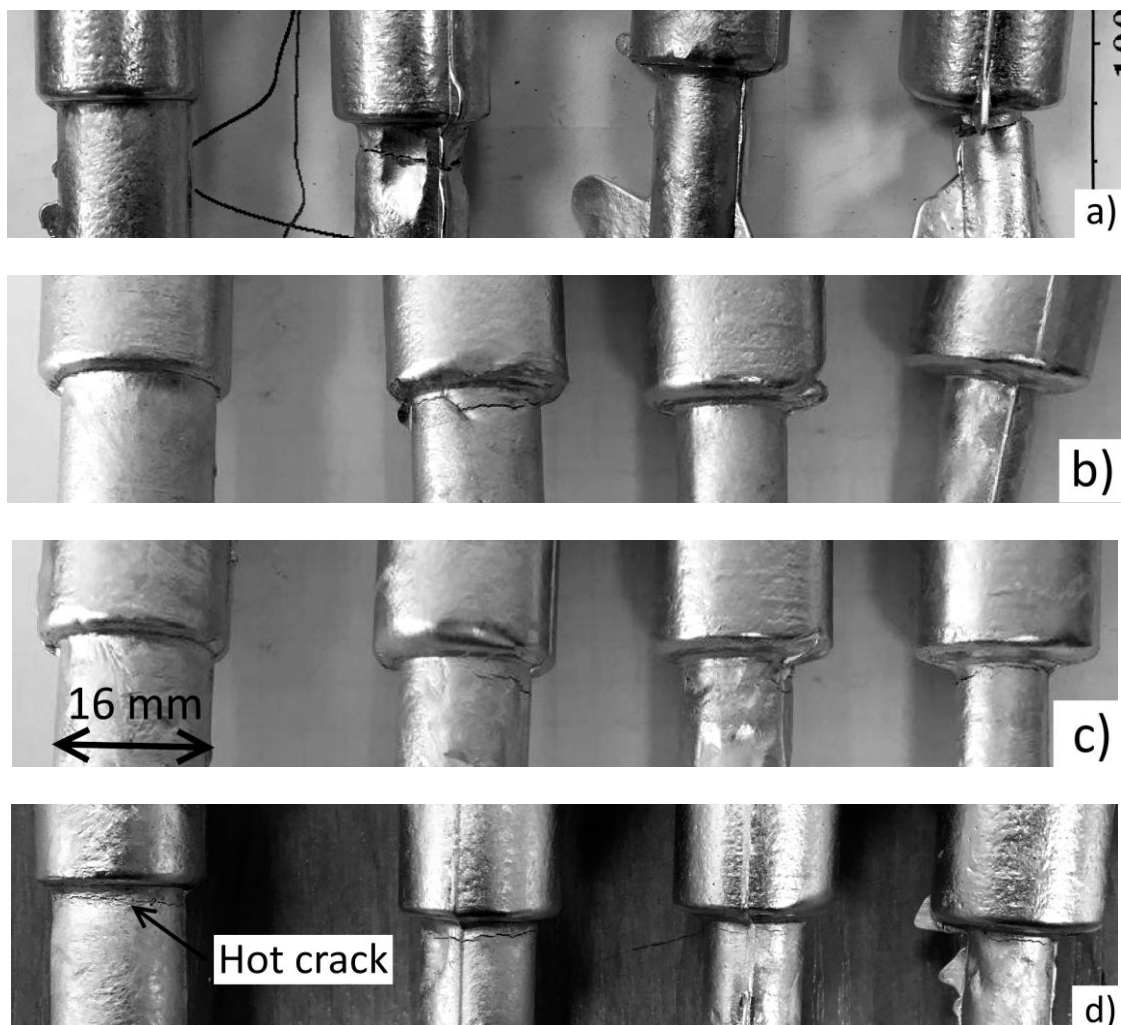


Figure 2 Pencil-type hot tearing test castings:
(a) A1 alloy; (b) A2 alloy; (c) A3 alloy; (d) A206 commercial alloy

The appropriate casting properties provided by specific solidification may not be achieved at lower solidification rate. **Figure 3** is dedicated to microstructures after slow cooling. They contain a great many needle like phases which are most likely to be Al_3Fe phase. Meanwhile, at similar conditions A1 alloy predominantly consists of needle-like constituents of different color (**Figure 3a**). Some of them are probably Al_3Ni phase. Besides needle phase, A2 alloy with calcium contains also a fishbone type phase which is most likely to be $(Al,Zn)_4Ca$ (**Figure 3b**). A3 alloy looks a bit finer due to specific Al_3Fe phase shape. They are predominantly separated as star shaped inclusions of about 20–30 μm of secondary crystallization origin. These structures looks unfavorable to obtain high ductile and due to this fact the new promising alloys are to be produced by permanent mold casting. But we assume that the same alloying approach may be effective to develop alloys for sand molds casting too.

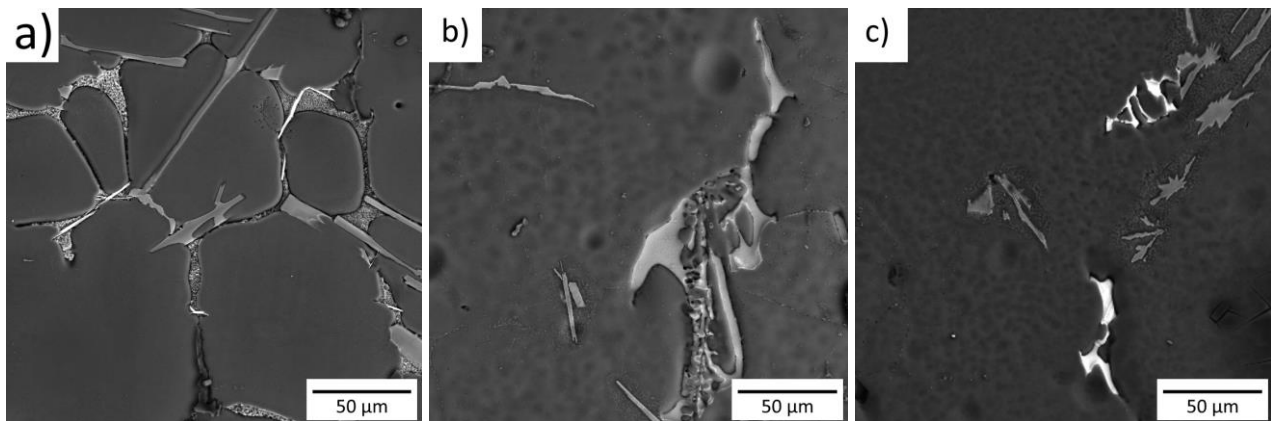


Figure 3 As-cast microstructures of experimental alloys after slow cooling:
(a) A1 alloy; (b) A2 alloy; (c) A3 alloy

Microstructures of the permanent mold castings are much finer. **Figure 4a** shows a structure of the A1 nickel-containing alloy. Many studies show that this type of alloy is a great one to be subjected to spheroidizing heat treatment and produced as shape castings and wrought products with high ductility and strength. The presented structure contains enough rounded and small plates shape intermetallics with almost the same both content of nickel and iron that matches to Al_9FeNi phase. An aluminum solid solution is in supersaturated state despite the T phase presence. All eutectic constituents are in the dendritic cells boundaries and they mostly are not in a bonded state, that is able to contribute to ductile failure during tensile test. **Figure 4b** is dedicated to structure of the A2 calcium-containing alloy. There are separated eutectic constituents too there, but due to specific $(Al,Zn)_4Ca$ formation, it contains fishbone shape inclusions. It is a bit complicated to identify $Al_{10}CaFe_2$ phase, but the presence of iron in eutectic and lack of needle shaped phases indicates on its occurrence. The A3 cerium-containing alloy structure (**Figure 4c**) is comparable to those of the A1 alloy. Indication of iron-containing $Al_{10}CeFe_2$ phase is more apparent and the whole composition of the presented microprobe analysis result points on a small amount of T phase. Aluminum solid solution in the A2 and A3 samples is also supersaturated with almost the same amount of Zn and Mg as it was in (Al) of the A1 sample. This contributes to solid solution effect and enhances the strength due to aluminum lattice bending.

Eventually all the presented structures of experimental alloys as permanent mold castings look more favorable than structures of Al-Si alloys presented in **Figure 4d** ($Al_7Si_0.3Mg$ alloy) and **Figure 4e** ($Al_{12}Si_2Cu$ alloy). These reference alloys are extensively used in automotive industry. Even if $Al_7Si_0.3Mg$

could be modified with Sr and obtain fine structure it would need a heat treatment. And there are often a primary (Si) crystals formation occurs in Al₁₂Si₂Cu alloy.

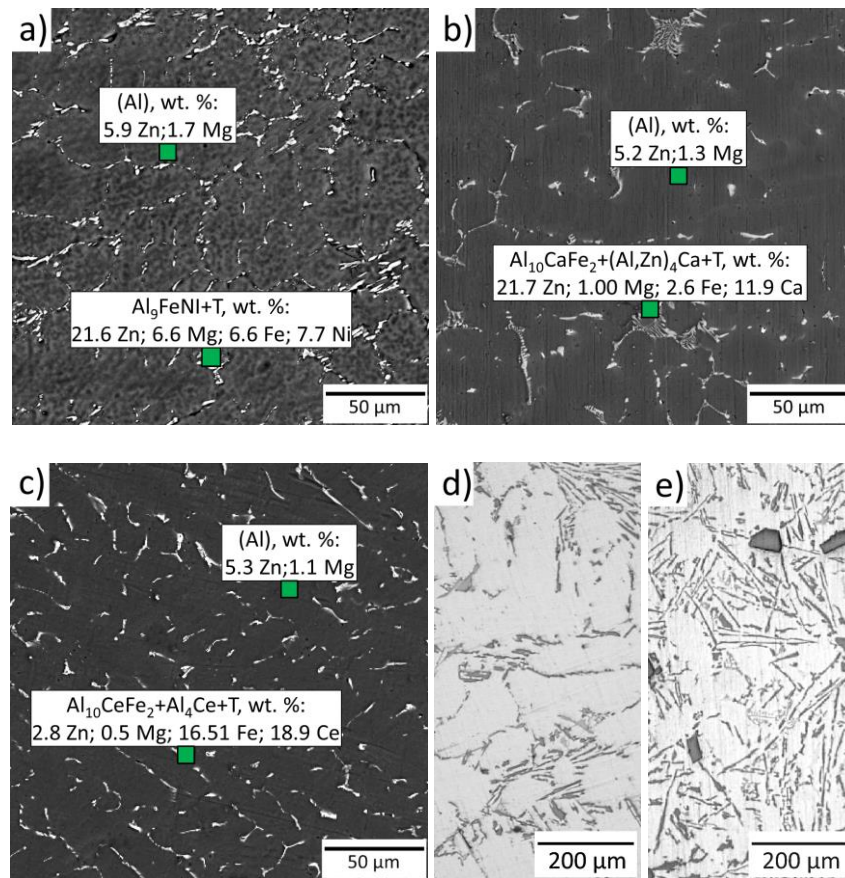


Figure 4 As-cast microstructures of experimental alloys after permanent mold casting:

(a) A1 alloy; (b) A2 alloy; (c) A3 alloy; (d) Al₇Si_{0.3}Mg alloy; (e) Al₁₂Si₂Cu alloy;

Figure 5 shows the tensile properties of experimental alloys under as-cast state. The properties of the A1 and A2 alloys are similar. UTS value of more than 330 MPa and YS value of more than 220 MPa have been achieved. The elongation is more than 5 %. The A3 alloy has a great reserve to be strengthened due to low YS value and the highest elongation. The whole results show a great advantage of the new promising alloys over commercial non-heat-treatable casting aluminum alloys.

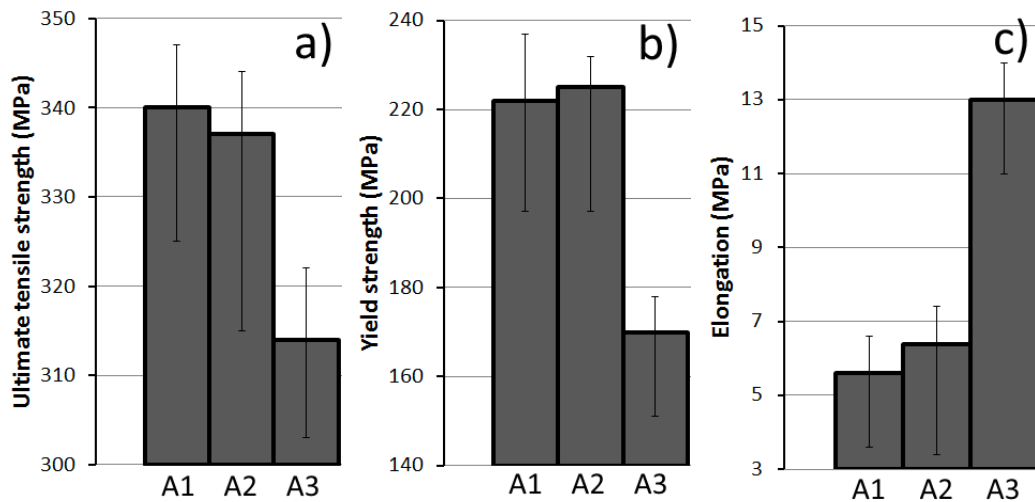


Figure 5 Tensile properties of experimental alloys under as-cast state: (a) UTS; (b) YS; (c) EI

4. CONCLUSION

New generation promising casting aluminum alloys was studied under as-cast state. The efficiency of eutectic-forming elements from Ni, Ca, Ce group and iron in hot tearing resistance enhance was substantiated. All the experimental alloys had better casting properties than those of A206 commercial alloy. The opportunity to obtain fine dispersed structure contained iron-containing ternary phases Al_9FeNi , $Al_{10}CaFe_2$ and $Al_{10}CeFe_2$ after permanent mold casting was confirmed. Both the intermetallics of eutectic origin distribution and supersaturated solid solution provide high mechanical properties which are exceeding those of existing commercial casting non-heat-treatable aluminum alloys. Moreover high content of iron provide the opportunity to obtain them using secondary stock or low-grade pure aluminum.

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