Reduction of the monounsaturated fatty acid content of *Escherichia coli* results in increased resistance to oxidative damage

Gonzalo A. Pradenas, Braulio A. Paillavil, Sebastián Reyes-Cerpa, José M. Pérez-Donoso and Claudio C. Vásquez

Departamento de Biología, Facultad de Química y Biología, Universidad de Santiago de Chile, Santiago, Chile

Reactive oxygen species (ROSs) affect several macromolecules and cellular components in eukaryotic and prokaryotic cells. In this work, the effect of various ROS-generating compounds on the *Escherichia coli* membrane was studied. Membrane fatty acid profiles, oxidative damage levels and bacterial resistance to these toxicants were determined. Studies included wild-type cells as well as a strain exhibiting a modified monounsaturated fatty acid (MUFA) profile (accomplished by overexpressing the β -hydroxyacyl acyl carrier protein dehydratase-encoding gene, *fabA*). Levels of membrane MUFAs and oxidative damage markers decreased slightly upon toxicant exposure with a concomitant increase in cell resistance to these ROS-generating compounds. A direct relationship between MUFAs and lipid peroxidation was observed. The lower the MUFA the lower the peroxide levels, suggesting that MUFAs are targets for membrane lipid oxidation.

Correspondence Claudio C. Vásquez claudio.vasquez@usach.cl

Received9 November 2011Revised27 January 2012Accepted10 February 2012

INTRODUCTION

It is well established that reactive oxygen species (ROSs) are generated as byproducts during the normal functioning of the electron transport chain (Imlay, 2003). ROSs are highly reactive and can damage several cell macromolecules such as proteins, nucleic acids and lipids, among others (Ercal *et al.*, 2001). In addition, ROSs such as hydrogen peroxide (H₂O₂), superoxide (O₂⁻) and hydroxyl radical (OH[•]) can also be formed during cell exposure to redox-cycling toxicants like paraquat (Bus & Gibson, 1984) and metals (iron) or metalloid salts such as chromate and potassium tellurite (Ercal *et al.*, 2001; Bagchi *et al.*, 2002; Pérez *et al.*, 2007).

Although it has been largely thought that the *Escherichia coli* membrane fatty acids are virtually resistant to ROS-induced injury (Imlay, 2003), evidence for oxidative damage to these membrane components has become available recently. It has been shown that *E. coli* lacking the oxidative stress defence regulator *oxyR* exhibits increased levels of thiobarbituric acid-responsive substances (TBARS, a common lipid pero-xidation marker) when exposed to *tert*-butyl-hydroperoxide (TBH) (Yoon *et al.*, 2002). This behaviour has also been observed in bacteria exposed to different oxidants such as

A supplementary figure is available with the online version of this paper.

hydrogen peroxide, potassium tellurite (K₂TeO₃), paraquat or TiO₂, among others (Maness *et al.*, 1999; Semchyshyn *et al.*, 2005; Pérez *et al.*, 2007, 2008).

We were interested in studying the E. coli response to oxidants, and we found that E. coli grown under conditions that resulted in changes to the membrane monounsaturated fatty acid (MUFA) ratio exhibited increased levels of TBARS, suggesting MUFA involvement in lipid peroxidation events. To further test this observation, we modified the MUFA ratio in the E. coli membrane by overexpressing fabA, encoding the β -hydroxyacyl acyl carrier protein dehydratase. This enzyme, involved in MUFA biosynthesis (Birge & Vagelos, 1972), is known to decrease the membrane MUFA levels when overproduced (Clark et al., 1983). In this work, we found that overexpressing fabA, besides rendering E. coli more resistant to oxidative stress (particularly to membrane-oxidizing compounds), results in decreased lipid-derived peroxides, cell oxidative status and oxidized proteins levels.

METHODS

Plasmids and culture conditions. Plasmid p*fabA* was constructed by ligating the amplified *E. coli fabA* gene into the pBAD TOPO (Invitrogen) expression vector according to the manufacturer's specifications. Primers used to amplify *fabA* were 5'-ATGGTAGATAAACG-CGAATCC-3' (forward) and 5'-GAAGGCAGACGTATCCTGGA-3' (reverse). The resulting p*fabA* was introduced into *E. coli* Top10 by electroporation. Correct *fabA* insertion was confirmed by *NcoI* digestion and PCR amplification. Then pBAD (control) and p*fabA*

Abbreviations: FOX, ferrous oxidation in the presence of xylenol orange; MUFA, monounsaturated fatty acid; ROS, reactive oxygen species; TBARS, thiobarbituric acid-responsive substances; TBH, *tert*-butylhydroperoxide.

Table 1. Fatty acid profile (%) of E. coli BW25113 exposed to different toxicants

Fatty acids from *E. coli* carrying pBAD or p*fabA* plasmids were subjected to direct *trans*-esterification and analysed by gas chromatography. Results are expressed as the percentage of total fatty acids. C16:1, palmitoleic acid; C16, palmitic acid; C18:1, vaccenic acid; Δ % C16:1+C18:1, variation of total MUFA content.

BW25113 pBAD	C16:1	C16	C18:1	C16:1+C18:1	Δ% C16:1+C18:1
Untreated	19.26 ± 0.81	39.70 ± 2.13	40.61 ± 0.83	59.86 ± 1.13	0
K ₂ TeO ₃	18.76 ± 0.98	43.76 ± 2.91	37.11 ± 0.99	55.84 ± 0.85	-4.15 ± 0.89
H_2O_2	18.72 ± 0.77	42.19 ± 2.16	38.78 ± 0.93	57.50 ± 1.08	-2.82 ± 0.50
TBH	17.48 ± 0.73	43.48 ± 1.42	38.81 ± 0.65	56.30 ± 1.13	-4.10 ± 0.32
BW25113 pfabA					
Untreated	18.10 ± 0.31	63.89 ± 1.72	17.18 ± 0.83	35.28 ± 0.88	0
K ₂ TeO ₃	16.05 ± 0.73	67.87 ± 1.71	15.08 ± 0.54	31.13 ± 0.67	-4.16 ± 1.51
H_2O_2	16.71 ± 1.20	67.72 ± 1.57	17.03 ± 1.12	33.77 ± 0.68	-1.51 ± 1.06
TBH	16.55 ± 0.54	67.05 ± 0.96	15.92 ± 1.08	32.47 ± 1.45	-2.81 ± 0.58

plasmids were introduced independently into *E. coli* BW25113. Cells were routinely grown in Vogel-Bonner (VB) minimal medium (Vogel & Bonner, 1956) containing 100 μ g ampicillin ml⁻¹ (Amp-VB).

MIC determination. Cells from overnight cultures were diluted 100fold with fresh Amp-VB medium and grown to OD_{600} 0.3. Then, 10 µl was inoculated into 96-well plates containing serial dilutions of the toxicant in 150 µl VB medium. Plates were incubated for 48 h at 37 °C with shaking. ROS elicitors tested included the superoxidegenerating compound tellurite (Pérez *et al.*, 2007), the lipidperoxidating agent TBH (Masaki *et al.*, 1989) and hydrogen peroxide (Imlay, 2003).

Fatty acid profile determination. *E. coli* BW25113 cultures (20 ml) treated for 30 min with K_2 TeO₃ (0.5 μ M), H_2O_2 (300 μ M) or TBH (50 μ M) were sedimented and dehydrated by using a Speed Vac. Dehydrated cells were suspended in 500 μ l methanolic HCl (5%) and incubated at 80 °C for 10 min to form fatty acid methyl esters. The mixture was extracted with hexane (500 μ l) and evaporated to a volume of 50 μ l. An aliquot (5 μ l) of this solution was injected in a diphenyl-dimethylsiloxane (5:95) column (PTE-5 Supelco) and fractionated by using a Perkin Elmer Autosystem gas chromatograph apparatus.

Protein carbonylation. Protein carbonyl group content was determined as described previously (Semchyshyn *et al.*, 2005). Briefly, nucleic acid-free cell extracts (100 µl) prepared from *E. coli* BW25113 exposed for 30 min to K₂TeO₃ (0.5 µM), H₂O₂ (300 µM) or TBH (50 µM) were mixed with 4 vols 10 mM 2,4-dinitrophenilhydrazine (2,4-DNHP; dissolved in 2 M HCl) and incubated for 1 h at room temperature with agitation. Proteins were precipitated with 1 vol 20 % trichloroacetic acid and sedimented at 10 000 *g* for 10 min. After washing three times with ethanol/ethyl acetate (1:1)

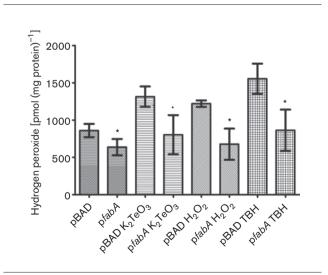
Table 2. MIC (μ M) of the indicated toxicants for *E. coli* carrying pBAD or p*fabA*

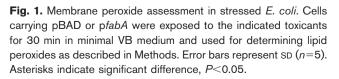
MICs were determined in VB minimal medium as described in Methods.

E. coli	K ₂ TeO ₃	H_2O_2	TBH
BW25113 pBAD	1.25	600	125
BW25113 pfabA	2.50	2500	500

the pellet was dissolved in 450 μ l 6 M guanidine hydrochloride containing 50 mM dithiothreitol. Carbonyl group content was determined at 370 nm using a molar absorption coefficient of 22 000 M⁻¹ cm⁻¹ (Johnson, 1953).

Lipoperoxide content determination. Lipid peroxidation levels were assessed using the FOX (ferrous oxidation in the presence of xylenol orange) assay as described by Cha *et al.* (2004) with some modifications. Briefly, *E. coli* (50 mg fresh cell paste) was suspended in 500 μ l Tris/HCl (pH 7.4) buffer, disrupted by sonication and centrifuged at 10 000 *g* for 10 min. Both protein extracts and pellets were sonicated again, this time in the presence of 1% SDS and extracted twice with 1 vol methanol/chloroform (2:1, v/v). The organic phase was recovered, evaporated in a SpeedVac and suspended in 1 ml FOX reagent. After 1 h at room temperature, the lipoperoxide content was determined at 560 nm. Results were expressed as hydrogen peroxide equivalents.





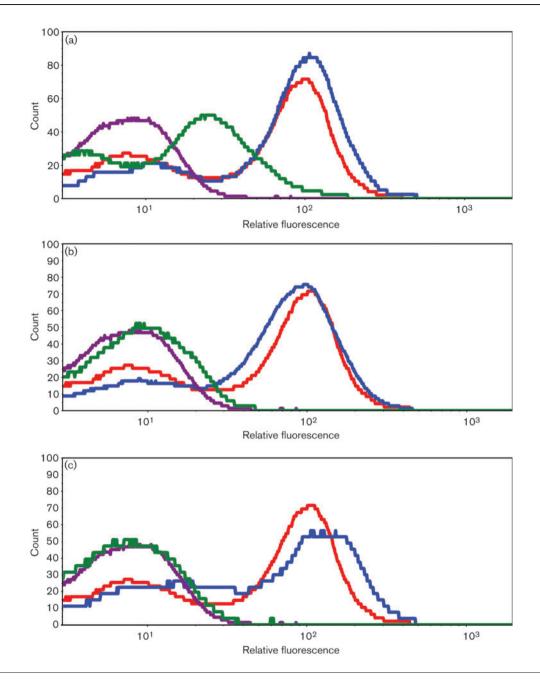


Fig. 2. Oxidative status assessment using the H₂DCFDA probe in toxicant-exposed *E. coli*. Flow cytometry analysis was carried out as described in Methods. *E. coli* carrying pBAD or pfabA was exposed to K₂TeO₃ (a), H₂O₂ (b) or TBH (c). A representative experiment is shown. Red, pBAD untreated; blue, pBAD treated; purple, pfabA untreated; green, pfabA treated.

Intracellular oxidative status. Intracellular ROS was assessed using the oxidation-sensitive probe 2',7'-dihydrodichlorofluorescein diacetate (H₂DCFDA) (Royall & Ischiropoulos, 1993), which is activated by several intracellular oxidants including hydrogen peroxide and the hydroxyl radical, among others. H₂DCFDA has also been successfully used to detect lipid peroxides and peroxyl radicals (Aoyagi *et al.*, 1999). Cells grown to OD₆₀₀ 0.3 were exposed for 30 min to potassium tellurite (0.5 μ M), hydrogen peroxide (300 μ M) or TBH (50 μ M). Then cells were washed twice, diluted 50-fold with PBS buffer and treated for 30 min with H₂DCFDA. Cells were washed once more and analysed with a Becton Dickinson FacsCanto II cytometer (Argon laser, detection at 520 nm).

RESULTS AND DISCUSSION

Owing to the virtual unreactivity of MUFA with superoxide radicals (Bielski *et al.*, 1983), it has been generally accepted that the *E. coli* membrane is rather inert to ROS. However, this assumption is supported by studies performed in conditions far from those found in biological systems and does not consider that MUFAs can generate lipid peroxides and oxidation-derived aldehydes (Loidl-Stahlhofen *et al.*, 1995). Additional evidence shows that MUFAs can react with ROS to form toxic compounds in eukaryotes (Borchman & Yappert, 1998; Rontani, 1998). In this context, our strategy to determine the effects of ROSgenerating compounds on the *E. coli* membrane was to modify the membrane fatty acid composition. To accomplish this, the *E. coli fabA* gene was amplified, cloned and expressed in this bacterium.

To assess the effect of fabA expression on the lipid composition of the E. coli membrane, its fatty acid content was analysed by gas chromatography (GC). Higher amounts of palmitic (C16), palmitoleic (C16:1) and vaccenic (C18:1) acids were detected in membrane samples. Since it was found that <2% of the cyclopropane comprised palmitoleic acid, it was not considered in the analysis depicted in Table 1. As expected, fabA expression resulted in augmented saturated fatty acid content (e.g. palmitic acid), as well as in decreased MUFA proportions (Table 1). Next, and to determine cell susceptibility to the toxicants, MICs of TeO_3^{2-} , H_2O_2 and TBH were determined for E. coli BW25113 carrying the plasmids pBAD (control) or pfabA. Upon fabA overexpression, cell resistance to tellurite and hydrogen peroxide or TBH increased by twoand fourfold, respectively (Table 2), suggesting an inverse relationship between MUFA content and resistance to oxidative stress. Such a relationship also suggests that MUFAs could be targets of oxidative damage that result in the generation of toxic derivatives such as aldehydes and lipid peroxides. No significant differences were observed in control and fabA-expressing cells regarding growth and resistance to Ni²⁺, a metal whose toxicity does not involve direct generation of ROS (Macomber & Hausinger, 2011) (data not shown). In addition, the E. coli membrane fatty acid profile was analysed in cells exposed to oxidative stress elicitors. E. coli carrying pBAD or pfabA exhibited a slightly lower unsaturated fatty acid (C16:1+C18:1) content when exposed to ROS-generating compounds (Table 1). In this context, it is interesting to note that the foremost effect was observed following tellurite treatment; MUFA levels decreased equally following toxicant treatment, irrespective of fabA expression.

Since the membrane MUFA content may be affected as a consequence of toxicant-induced lipid oxidation, membrane peroxidation was evaluated by using the FOX assay. Control E. coli had higher amounts of peroxide than fabAexpressing cells in all conditions (Fig. 1) apart from NiCl₂ treatment (not shown). Similar results were obtained regarding TBARS levels (Fig. S1, available with the online version of this paper). These results suggest a direct relationship between MUFA content and lipid peroxide generation. Monounsaturated fatty acids would be ROS targets, the oxidation of which would produce lipid peroxides in the same manner in which oleic acid (a MUFA) is oxidized to form hydroperoxides (Loidl-Stahlhofen et al., 1995). Curiously, fabA expression results in significant lipid peroxidation even in the control condition; this most probably occurs because of the continuous ROS production during the normal functioning of the respiratory chain (Messner & Imlay, 1999). Thus,

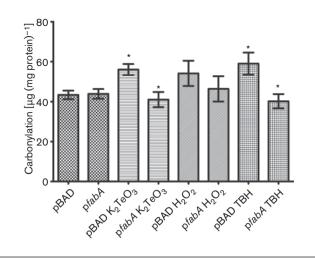


Fig. 3. Protein carbonylation in *E. coli* exposed to different toxicants. Cells carrying pBAD or p*fabA* were exposed to the indicated oxidative stress elicitors for 30 min in minimal media, and carbonyl groups were assessed using 2,4 DNPH as indicated in Methods. Error bars represent SD (n=5). Asterisks indicate significant difference, P<0.05.

the lower the MUFA content, the lower the lipid peroxidation levels.

Since lipid oxidation byproducts - as peroxides and/or reactive aldehydes - cause secondary oxidative damage to several macromolecules (Stadtman & Levine, 2000), the intracellular oxidative status of E. coli exposed to the toxicants tested was evaluated. In this context, intracellular levels of oxidant molecules were assessed by flow cytometry using the oxidation-sensitive probe H₂DCFDA. When exposed to the toxicants, pBAD-carrying cells showed little (tellurite and TBH) or no (H₂O₂) increase in probe activation compared with untreated controls. This low probe excitation may be explained because toxicant concentration was 1/4 of the respective MICs (see Methods for details). However, fabA-expressing E. coli showed decreased probe activation in treated or untreated conditions, in agreement with lower intracellular ROS levels (Fig. 2). These observations support the lipid oxidation results, and suggest that any event resulting in decreased MUFA levels actually diminishes the available substrates - including lipid peroxides - that can react with the fluorescent probe (Rontani, 1998).

Another way by which oxidants damage cell components is through carbonylation events that affect proteins (Yan, 2009). This chemical modification is commonly used as a marker to assess protein oxidation. While exposure to oxidative damage elicitors increased protein carbonylation levels, *fabA* overexpression resulted in decreased protein oxidation in cells treated with potassium tellurite or TBH (Fig. 3). These results are consistent with those in which lipid peroxides were determined (Fig. 1) and suggest that both events could be membrane-related, since carbonylation is caused by membrane-derived oxidized products such as TBARS (Esterbauer *et al.*, 1991).

In summary, our results suggest that the *E. coli* membrane composition is relevant in determining the oxidative status of the cell as well as in resistance to oxidative stress elicitors. In this context, MUFAs could react with ROS or ROS-generating compounds, thus producing oxidized membrane degradation byproducts (Loidl-Stahlhofen *et al.*, 1995). However, additional experiments with cells exhibiting altered membrane composition are needed to get a more complete picture of the relevance of membranes in *E. coli* facing oxidative stress.

ACKNOWLEDGEMENTS

This work was supported in part by FONDECYT (Fondo Nacional de Investigación Científica y Tecnológica) grant nos 1090097 (C. C. V.) and 3100049 (J. M. P.). Dicyt (Dirección de Investigación)-USACH (C. C. V.) and 021043PD (J. M. P.) and IFS (International Foundation for Science) grant no. F/4733 (J. M. P.) are also acknowledged. G. A. P. was supported by a doctoral fellowship from MECESUP (Mejoramiento de la Calidad y Equidad de la Educación Superior) UCH106, Chile. B. A. P. and S. R.-C. received doctoral fellowships from CONICYT (Comisión Nacional de Ciencia y Tecnología), Chile.

REFERENCES

Aoyagi, K., Akiyama, K., Tomida, C., Gotoh, M., Hirayama, A., Takemura, K., Ueda, A., Nagase, S., Koyama, A. & Narita, M. (1999). Imaging of hydroperoxides in a rat glomerulus stimulated by puromycin aminonucleoside. *Kidney Int* Suppl 71, 153–155.

Bagchi, D., Stohs, S. J., Downs, B. W., Bagchi, M. & Preuss, H. G. (2002). Cytotoxicity and oxidative mechanisms of different forms of chromium. *Toxicology* 180, 5–22.

Bielski, B. H., Arudi, R. L. & Sutherland, M. W. (1983). A study of the reactivity of HO_2/O_2^- with unsaturated fatty acids. *J Biol Chem* **258**, 4759–4761.

Birge, C. H. & Vagelos, P. R. (1972). Acyl carrier protein. XVII. Purification and properties of -hydroxyacyl acyl carrier protein dehydrase. *J Biol Chem* **247**, 4930–4938.

Borchman, D. & Yappert, M. C. (1998). Age-related lipid oxidation in human lenses. *Invest Ophthalmol Vis Sci* 39, 1053–1058.

Bus, J. S. & Gibson, J. E. (1984). Paraquat: model for oxidantinitiated toxicity. *Environ Health Perspect* 55, 37–46.

Cha, M. K., Kim, W. C., Lim, C. J., Kim, K. & Kim, I. H. (2004). *Escherichia coli* periplasmic thiol peroxidase acts as lipid hydroperoxide peroxidase and the principal antioxidative function during anaerobic growth. *J Biol Chem* 279, 8769–8778.

Clark, D. P., DeMendoza, D., Polacco, M. L. & Cronan, J. E., Jr (1983). Beta-hydroxydecanoyl thio ester dehydrase does not catalyze a ratelimiting step in *Escherichia coli* unsaturated fatty acid synthesis. *Biochemistry* 22, 5897–5902.

Ercal, N., Gurer-Orhan, H. & Aykin-Burns, N. (2001). Toxic metals and oxidative stress part I: mechanisms involved in metal-induced oxidative damage. *Curr Top Med Chem* 1, 529–539. Esterbauer, H., Schaur, R. J. & Zollner, H. (1991). Chemistry and biochemistry of 4-hydroxynonenal, malonaldehyde and related aldehydes. *Free Radic Biol Med* 11, 81–128.

Imlay, J. A. (2003). Pathways of oxidative damage. *Annu Rev Microbiol* 57, 395–418.

Johnson, G. D. (1953). Correlation of color and constitution. I. 2, 4dinitrophenylhydrazones. *J Am Chem Soc* 75, 2720–2723.

Loidl-Stahlhofen, A., Kern, W. & Spiteller, G. (1995). Gas chromatographic-electron impact mass spectrometric screening procedure for unknown hydroxyaldehydic lipid peroxidation products after pentafluorobenzyloxime derivatization. *J Chromatogr B Biomed Appl* **673**, 1–14.

Macomber, L. & Hausinger, R. P. (2011). Mechanisms of nickel toxicity in microorganisms. *Metallomics* **3**, 1153–1162.

Maness, P. C., Smolinski, S., Blake, D. M., Huang, Z., Wolfrum, E. J. & Jacoby, W. A. (1999). Bactericidal activity of photocatalytic TiO₂ reaction: toward an understanding of its killing mechanism. *Appl Environ Microbiol* 65, 4094–4098.

Masaki, N., Kyle, M. E. & Farber, J. L. (1989). *tert*-Butyl hydroperoxide kills cultured hepatocytes by peroxidizing membrane lipids. *Arch Biochem Biophys* 269, 390–399.

Messner, K. R. & Imlay, J. A. (1999). The identification of primary sites of superoxide and hydrogen peroxide formation in the aerobic respiratory chain and sulfite reductase complex of *Escherichia coli*. *J Biol Chem* **274**, 10119–10128.

Pérez, J. M., Calderón, I. L., Arenas, F. A., Fuentes, D. E., Pradenas, G. A., Fuentes, E. L., Sandoval, J. M., Castro, M. E., Elías, A. O. & Vásquez, C. C. (2007). Bacterial toxicity of potassium tellurite: unveiling an ancient enigma. *PLoS ONE* 2, e211.

Pérez, J. M., Arenas, F. A., Pradenas, G. A., Sandoval, J. M. & Vásquez, C. C. (2008). *Escherichia coli* YqhD exhibits aldehyde reductase activity and protects from the harmful effect of lipid peroxidation-derived aldehydes. *J Biol Chem* 283, 7346–7353.

Rontani, J. F. (1998). Photodegradation of unsaturated fatty acids in senescent cells of phytoplankton: photoproduct structural identification and mechanistic aspects. *J Photochem Photobiol Chem* **114**, 37–44.

Royall, J. A. & Ischiropoulos, H. (1993). Evaluation of 2',7'-dichlorofluorescin and dihydrorhodamine 123 as fluorescent probes for intracellular H_2O_2 in cultured endothelial cells. *Arch Biochem Biophys* **302**, 348–355.

Semchyshyn, H., Bagnyukova, T., Storey, K. & Lushchak, V. (2005). Hydrogen peroxide increases the activities of *soxRS* regulon enzymes and the levels of oxidized proteins and lipids in *Escherichia coli*. *Cell Biol Int* 29, 898–902.

Stadtman, E. R. & Levine, R. L. (2000). Protein oxidation. *Ann N Y Acad Sci* **899**, 191–208.

Vogel, H. J. & Bonner, D. M. (1956). Acetylornithinase of *Escherichia coli:* partial purification and some properties. *J Biol Chem* **218**, 97–106.

Yan, L. J. (2009). Analysis of oxidative modification of proteins. *Curr Protoc Protein Sci* Chapter 14, t14, 4.

Yoon, S. J., Park, J. E., Yang, J. H. & Park, J. W. (2002). OxyR regulon controls lipid peroxidation-mediated oxidative stress in *Escherichia coli*. J Biochem Mol Biol **35**, 297–301.

Edited by: G. H. Thomas