1400059





North American Branch 32 Avenue of the Americas New York, NY 10013-2473, USA

Tel :	212 337-5978
Fax :	212 337-5959
E-mail:	mgillette@cambridge.org

Dear Microscopy and Microanalysis Contributor:

**PROOFREADING INSTRUCTIONS** 

Attached is a PDF page proof of your introduction/article/book review scheduled to be published in:

# Microscopy and Microanalysis

Please follow these procedures:

- 1. **Proofreading**: Proofread your article carefully. Check especially the spellings of names and places as well as the accuracy of dates and numbers. Please answer all queries that may appear on a separate page.
- 2. **Text**: Changes in the text are limited to typographical and factual errors. Rewriting or other stylistic changes are not permitted. Contributors may be charged for excessive author alterations, and publication of your article may be delayed.
- 3. **Corrections**: Please respond with an e-mail message to Morrell Gillette at mgillette@cambridge.org, identifying the correction by page number, column, paragraph, and line. Please indicate the present errant copy followed by the correct copy. The corrections to the proofs should be sent within 2 days of receipt. Corrections can also be entered into the PDF as embedded comments.
- 4. **References**: If the query involves a reference entry, please arrange the new entry into the correct format. In order to prepare the HTML full-text files for online viewing, the complete information including correct spelling of author names, year, titles, publisher, city of publication, page range, and so forth are needed.
- 5. **Figures**: Review the figure reproductions on the page proofs to see if important features have been well represented. If something seems out of order, indicate the errant features in a cover letter. Should it be necessary that new electronic copy of the figures (in PDF, TIFF, or EPS) or text (Word or LaTeX) will have to be provided, please indicate which application is being used.
- 6. **Offprints or Bound Copies**: Free offprints are longer being offered. However, a message with a link to access a free PDF of your paper will be sent to you. To order reprints or offprints of your article or printed copies of the issue, please visit the Cambridge University Reprint Order Center online at: www.sheridan.com/cup/eoc
- 7. **Delay in response**: Please note that failure to respond in a timely fashion may delay publication of your article or may require publication without your corrections. Thank you for your prompt attention to these proofs. If you have any questions, please feel free to contact Morrell Gillette by e-mail: mgillette@cambridge.org, or at (212) 337-5978. Thanks.

Best Regards, Morrell Gillette, Production Editor 7

# QUERY FORM

MAM		
Manuscript ID	[Art. Id: 1400059]	
Author		
Editor		
Publisher		

## Journal: Microscopy And Microanalysis

Author :- The following queries have arisen during the editing of your manuscript. Please answer queries by making the requisite corrections at the appropriate positions in the text.

Query No	Nature of Query
Q1	The distinction between surnames can be ambiguous, therefore to ensure accurate tagging for indexing purposes online (eg for PubMed entries), please check that the highlighted surnames have been correctly identified, that all names are in the correct order and spelt correctly.
Q2	Please provide manufacturer details for Hitachi, Gatan Enfina, Nion UltraSTEM.
Q3	Please provide volume number and page range in Ref. Carretero-Genevrier et al. (2013).
Q4	Please provide last page number in Ref. Luo et al. (2009), if applicable.

1



Q1 5 Jaume Gazquez,<sup>1,\*</sup> Adrián Carretero-Genevrier,<sup>2,3</sup> Martí Gich,<sup>1</sup> Narcís Mestres,<sup>1</sup> and María Varela<sup>4,5</sup>

<sup>1</sup>Institut de Ciència de Materials de Barcelona ICMAB, Consejo Superior de Investigaciones Científicas CSIC, 08193 Bellaterra,
 7 Spain

8 <sup>2</sup>Laboratoire Chimie de la Matière Condensée, UMR UPMC-Collège de France-CNRS 7574, Collège de France, 11 Place

9 Marcelin Berthelot, 75231 Paris, France

<sup>3</sup>Institut des Nanotechnologies de Lyon (INL) CNRS - Ecole Centrale de Lyon., 36 avenue Guy de Collongue, 69134 Ecully

<sup>4</sup>*Materials Science and Technology Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831, USA* 

12 <sup>5</sup>Departamento de Física Aplicada III & Instituto Pluridisciplinar, Universidad Complutense de Madrid, Madrid 28040, Spain

13 **Abstract:** In this study we combine scanning transmission electron microscopy, electron energy loss 14 spectroscopy and electron magnetic circular dichroism to get new insights into the electronic and magnetic 15 structure of LaSr-2×4 manganese oxide molecular sieve nanowires integrated on a silicon substrate. These 16 nanowires exhibit ferromagnetism with strongly enhanced Curie temperature ( $T_c > 500$  K), and we show that the 17 new crystallographic structure of these LaSr-2×4 nanowires involves spin orbital coupling and a mixed-valence 18 Mn<sup>3+</sup>/Mn<sup>4+</sup>, which is a must for ferromagnetic ordering to appear, in line with the standard double exchange

19 explanation.

20 Key words: EELS, STEM, nanoparticles, nanowires, manganese molecular sieves, white lines, oxidation state

#### 

Magnetic oxide nanoparticles are interesting for a wide range 22 of novel devices since they constitute the building blocks for 23 future data storage devices or magnetic sensors. However, for 24 practical device applications, magnetic nanomaterials should 25 26 exhibit a  $T_c$  well above room temperature. Many complex oxides, such as mixed-valence perovskite manganites, diluted 27 magnetic oxides, spinels or ferrites, exhibit this requirement in 28 bulk. However, increasing  $T_c$  values above room temperature 29 in nano-systems remains a challenge, since the physical 30 properties of nanostructures often differ from those exhib-31 32 ited by their bulk counterparts. We have recently been able to synthesize manganese oxide molecular sieve nanowires that 33 display ferromagnetism above room temperature using a 34 sol-gel-based polymeric precursor solution and track-etched 35 polymer templates (Carretero-Genevrier et al., 2011, 2008). 36 Unfortunately, a microscopic explanation of the underlying 37 physics is lacking. This is, in part, due to the fact that the 38 magnetic characterization properties are based on macro-39 scopic measurements of a large number of nanostructures, 40 and one cannot relate the local structure of a single nanowire 41 with its actual averaged magnetic properties, so the nature of 42 magnetism in the nanometer scale is still unknown. 43

 In order to gain further knowledge in this front,
 real space probes with atomic resolution and sensitivity to
 magnetic quantities are needed. Among the experimental
 techniques available for characterization of nanostructured materials, aberration corrected scanning transmission electron microscopy (STEM), in combination with electron 48 energy loss spectroscopy (EELS), is a powerful technique for 49 structural and chemical analysis down to the atomic scale. 50 The STEM electron probe routinely yields spatial resolution 51 of < 1 Å as well as increased contrast both in imaging and in 52 spectroscopy (Varela et al., 2004, 2005), allowing simulta-53 neous mapping of the structure, chemistry, and electronic 54 properties of materials in a single experiment. Furthermore, 55 as has recently been demonstrated by Schattschneider et al. 56 (2006) electron probes can also be used to study the magnetic 57 properties. Energy-loss magnetic circular dichroism 58 (EMCD) can be measured in a TEM studying the  $L_{2,3}$  EELS 59 absorption edges of transition metal, and has already been 60 used to study nanostructured materials in the sub-61 nanometer range (Schattschneider et al., 2008; Zhang et al., 62 2009, 2011; Lidbaum et al., 2010; Carretero-Genevrier et al., 63 2012; Salafranca et al., 2012). 64

In this work, we will shed some light on the connection 65 between structure, chemistry, and magnetism in the nano-66 meter scale in manganese oxide molecular sieve nanowires. 67 Our LaSr-2  $\times$  4 nanowires present a  $T_c$  that is much higher 68 than most perovskite Mn oxides (above 500 K) so they can be 69 studied at room temperature in the aberration corrected 70 STEM. We will use atomic resolution STEM-EELS to 71 investigate the electronic properties of different Mn sites 72 within the monoclinic structure and combine them with 73 EMCD onto a single LaSr-2  $\times$  4 nanowire and attempt to 74 distinguishing the Mn spin and orbit contribution. Our 75 results will pave the way to elucidate the mechanism 76 underlying both ferromagnetism and the high Curie 77 temperature exhibited by these LaSr-2×4 nanowires. 78

## 79 MATERIALS AND METHODS

<sup>80</sup> LaSr-2×4 nanowires were prepared from 1 M aqueous solutions composed of La(NO<sub>3</sub>)3•6 H<sub>2</sub>O, Sr(NO3)2•4 H<sub>2</sub>O and Mn(NO<sub>3</sub>)2•4 H<sub>2</sub>O in the molar proportion of 07:0.3:1, respectively. The addition of ethylene glycol (EG) heated above 100°C promotes polymerization of the EG in order to reach the optimum viscosity value required for filling of the template's pores.

Track etched polymer templates were prepared by irra-87 diation of polyimide or polycarbonate directly supported on 88 single crystalline Si substrates by using heavy ions and pos-89 terior chemical development as described elsewhere (Carre-90 tero-Genevrier et al., 2010, 2011, 2012, 2013). Then, all 91 precursor solutions were used to fill the nanopores of the 92 polymer template. A thermal treatment at 800°C for 5 h 93 (temperature ramp 3°C min-1) in air was applied directly 94 after filling the nonporous template in a tubular furnace in 95 order to obtain vertical epitaxial LaSr-2×4 nanowires on a 96 silicon substrate.  $\varepsilon$ -MnO<sub>2</sub> nanowires up to 30  $\mu$ m in length, 97 98 were synthesized following the same procedure with a thermal treatment of 2 h at 700°C. These nanowires were 99 indexed with a hexagonal unit cell a = b = 5.28 Å, c = 2.86100 Å,  $\alpha = 90^\circ$ ,  $\beta = 90^\circ$  and  $\gamma = 120^\circ$ , described elsewhere 101 (Carretero-Genevrier et al., 2011). 102

The samples' microstructure was investigated using a field emission gun scanning electron microscope (FEG-205 SEM), Hitachi's SU77. A VG Microscope HB501UX and a Nion UltraSTEM, both operated at 100 kV and equipped with a Nion aberration corrector and a Gatan Enfina EEL spectrometer, were used for high-resolution imaging, 108 Z-contrast imaging, and STEM-EELS spectrum imaging. 109 The probe forming aperture had a semiangle of ~ 25 mrad. 110 The collection angle at the spectrometer was estimated to be 111 12 mrad. Thin foil cross-sectional samples of the wires grown 112 on the substrates were prepared for STEM by conventional 113 grinding and ion milling methods. The EMCD experiment 114 was also carried out using a Nion Ultrastem at 100 kV, but 115 operated in nanodiffraction mode. In this mode the scanning 116 beam has a diameter of about 1 nm. The convergence and 117 collection semiangles were set to 30 and 34 mrad, respec-118 tively. The detection of circular dichroism in the TEM is 119 described in the literature (Schattschneider et al., 2006). It 120 can be achieved by recording two EEL spectra in the dif-121 fraction plane with different *polarizations*, this is, having an 122 opposite sign of the imaginary part, I + and I -, by selecting 123 the scattering angles such that the momentum transfers q 124 and q' are orthogonal to each other,  $q \perp q'$ . 125

### **Results and Discussion**

Our nanowires are magnetic at room temperature. The 127 proposed unit cell of these nanowires is monoclinic with 128 lattice parameters a = 13.8 Å, b = 5.7 Å, c = 21.8 Å, and 129  $\beta = 101^{\circ}$ , where the long axis of the nanowires is along the b 130 crystallograghic direction (Carretero-Genevrier et al., 2008). 131 It also presents an ordered arrangement of the La<sup>3+</sup> and 132 Sr<sup>2+</sup> cations (Carretero-Genevrier et al., 2011). Figure 1a–1c 133 show a low magnification FEG-SEM image of the LaSr-2×4 134

126



**Figure 1. a**: Low magnification field emission gun scanning electron microscope (FEG-SEM) image of LaSr-2×4 nanowires epitaxially grown on a  $\alpha$ -quartz/Si substrate. **b**: Hysteresis loops of the LaSr-2×4 nanowires measured at 400 K for fields applied parallel (filled symbols), and perpendicular (open symbols), to the substrate plane. **c**: Hysteresis loops of the LaSr-2×4 nanowires measured at 10 K, showing a saturation magnetization of (Ms) of 47 ± 10 emu/g.

nanowires and the magnetic hysteresis loops measured for 135 fields up to 5 T, at 400 K and at 10 K, respectively. For the 136 400 K measurement, the magnetic field was applied in-plane 137 and out-of-plane with respect to the substrate. The out-of 138 plane magnetization curve saturates at a higher magnetic 139 field than does the in-plane curve, indicating that the direc-140 tion perpendicular to the long axis of the nanowires is a hard 141 magnetic axis. We have calculated the saturation magneti-142 zation value for the monoclinic LaSr-2×4 nanowires. The 143 mass of nanowires was estimated in  $2.86 \cdot 10^{-06}$  g, which gives 144 saturation magnetization of  $47 \pm 10$  emu/g for the LaSrа 145  $2 \times 4$  monoclinic phase at 10 K, which is half of the saturation 146 magnetization value of the La<sub>0.7</sub>Sr<sub>0.3</sub>MnO<sub>3</sub> bulk perovskite 147 (90 emu/g), indicating the high ferromagnetic strength of 148 this novel type of nanostructure. Considering the structure 149 proposed by Carretero-Genevrier et al. (2011), which is 150 similar to the RUB-7 manganese oxide structure (Rziha et al., 151 1996), one can estimate for the LaSr-2×4 nanowires a 152 1.8  $\mu_{\rm B}$ /Mn, slightly less than half of that of Mn<sup>3+</sup>, 4.9  $\mu_{\rm B}$ . 153

Figure 2a shows a typical cross-sectional low magnification high-angle annular dark field image STEM image of one of our samples. The brighter rods correspond to the randomly oriented LaSr-2×4 nanowires, which cover the 157 surface of a polycrystalline  $\alpha$ -quartz layer formed along with 158 the nanowire synthesis. Figure 2b is a high resolution TEM 159 image showing a LaSr-2×4 nanowire epitaxially grown on 160 top of an  $\alpha$ -quartz grain. The Fast Fourier transform (FFT) 161 of Figure 2b reveals that the orientation relationship can be 162 described as (010) LaSr-2×4//(010)  $\alpha$ -guartz and an in-163 plane epitaxial relationship given by [20-2] LaSr-2×4// 164 [-101]  $\alpha$ -quartz. STEM-EELS studies allowed us to explore 165 the chemistry of the nanowires. Figure 2c shows an EEL 166 spectrum generated from scanning a nanowire, confirming 167 its uniform composition. It is worth noting that neither 168 appreciable inter-diffusion of Si into the nanowires nor any 169 trace of Sr on the quartz layer was detected from our EELS 170 studies. 171

The crystal quality of the wires is high. The growth 172 process of these LaSr-2 × 4 nanowires on Si substrates has 173 been well established in recent works (Carretero-Genevrier 174 et al., 2008, 2011, 2013). The confinement imposed by a 175 polymer template during thermal treatment results in the 176 formation of  $\varepsilon$ -MnO<sub>2</sub> nanoparticles in the early stages of the 177 growth process. These act as seeds for the further growth of 178



**Figure 2. a**: Low magnification Z-contrast image of LaSr-2×4 nanowires epitaxially grown on an  $\alpha$ -quartz/Si substrate. The inset shows the FFT of both phases. **b**: High-resolution transmission electron microscopy (HRTEM) image of the  $\alpha$ -quartz-Si interface along [010]  $\alpha$ -quartz zone axis. **c**: Electron energy loss (EEL) spectrum generated from the EELS scans across a nanowire.



**Figure 3.** Mn  $L_{2,3}$  edges, after background subtraction, in the two configurations (+) and (-), and the resultant dichroic signal (upper and bottom panels, respectively). The spectra were normalized to the background region between the two edges. The inset shows the diffraction pattern and the detector positions used. The electron energy loss spectroscopy (EELS) spectrum has been corrected by the subtraction of a baseline (a double-step function) that takes into account a linear increase of the background inside the  $L_{2,3}$  edges, as well as the background out of the  $L_{2,3}$  edge. The red line is the integral for the signal. Blue arrows indicate the values of p and q.

the manganate nanowires at high temperature treatments 179 (i.e. 800°C). In addition, contact of the silica native layer with 180 alkaline-earth metal cations (like the Sr<sup>2+</sup> present in the 181 precursor solution) during the thermal treatment yields its 182 devitrification to an  $\alpha$ -quartz layer, as recently demonstrated 183 by the authors (Carretero-Genevrier et al., 2013). This is a 184 key factor for stabilization of these manganese oxide nano-185 wires. Indeed,  $\alpha$ -quartz film renders better lattice matching 186 to the complex oxide nanostructures favoring the epitaxial 187 growth of the LaSr-2×4 nanowires. However, due to the 188 relatively low annealing temperature (800°C), the  $\alpha$ -quartz 189 layer grown on Si substrates is polycrystalline, which results 190 in growth of the nanowires along different orientations with 191 respect to the substrate plane. 192

The magnetism of these nanowires can be studied using 193 EMCD, which, as mentioned before, can be measured in a 194 TEM studying  $L_{2,3}$  EELS absorption edges of transition 195 metals (Schattschneider et al., 2006). Following the proce-196 dures described by Schattschneider et al., we have obtained 197 the Mn-L edges from two conjugated spots in the nanodif-198 fraction diagram, which are equivalent to the two beam 199 polarizations used in X-ray magnetic circular dichroism 200 (XMCD) measurements (Fig. 3a). The dichroic signal is then 201 given by the difference between these two spectra. A clear 202 dichroism is observed in the LaSr-2×4 nanowire, as dis-203 played in the lower panel of Figure 3a. Further evidence of 204 the difference in Mn  $L_{2,3}$  edges comes from calculating the 205  $L_{2,3}$  ratio for each spectrum coming from the two conjugated 206 spots in the diffraction pattern. The value of the  $L_{2,3}$  ratio has 207 208 been obtained here using the second derivative method (Botton et al., 1995), giving the values 2.4 and 2.9 for posi-209 tions (+) and (-), respectively. EEL spectra were acquired 210

from relatively thin regions, with a thickness value in terms 211 of the inelastic mean-free path of  $t/\lambda < 0.5$ . 212

Quantification of the actual values of the spin and 213 orbital moments for the Mn ions is challenging. However, 214 the EMCD integrals might supply information about their 215 relative size and orientation. We have used two different 216 methods to calculate the quotient between the orbital  $(l_Z)$ 217 and spin  $(s_Z)$  moments projections: the approach followed by 218 Chen et al. for X-rays and that used by Calmels et al. (Chen 219 et al., 1995; Calmels et al., 2007), which takes into account 220 the dynamical diffraction effects arising from the propaga-221 tion of the electron diffracted beams within the crystal 222 (something that cannot be omitted in a quantitative analy-223 sis). First, we consider the use of the sum rules (Chen et al., 224 1995), which may provide the relative size and orientation of 225 the  $s_Z$  and  $l_Z$  moments of Mn. Figure 3b displays the result of 226 applying sum rules to Mn ions on the LaSr-2×4 nanowires. 227 Bear in mind that the only XMCD expression that can be 228 applied in our experiment is for the ratio of the orbital and 229 the spin moment. q and p are the values of the integrated 230 areas of the EMCD signal of the Mn  $L_{2,3}$  edges and the  $L_3$ 231 edge, respectively (Fig. 3b). The obtained value for the 232 quotient  $l_Z/s_Z$  is -0.82. The negative value points to an 233 antiparallel alignment of the spin and the orbital magnetic 234 moments. 235

These results can be compared with those obtained 236 using the sum rules derived for EMCD spectra, which take 237 into account dynamical effect, sample orientation, and 238 thickness (Calmels et al., 2007). Following the procedure 239 described by Calmels et al., (2007) and having integrated the 240 EMCD spectrum in the energy windows [632 eV, 644 eV] for 241 the  $L_3$  edge and [646 eV, 657 eV] for the  $L_2$  edge, the mea-242 surement then gives  $l_Z/s_Z = -0.62$ . This result is in agree-243 ment with the value previously obtained. 244

In order to get new insights into the mechanism 245 responsible for the high temperature ferromagnetism we 246 studied and compared the electronic structure of these LaSr-247  $2 \times 4$  nanowires with that of the  $\varepsilon$ -MnO<sub>2</sub> phase, which 248 is nonmagnetic, has a unique Mn valence value and, as 249 mentioned before, serves as a seed for the growth of the 250 manganite nanowires. In manganites, the small peak in the 251 very onset of the O-K edge is related to the hybridization of 252 the O 2p band and the Mn 3d states, and it is found to change 253 with manganese oxidation state (Varela et al., 2009). In 254 addition, as confirmed by several studies, the ratio between 255  $L_3$  and the  $L_2$  of the  $L_{2,3}$  edge also changes as a function of the 256 occupancy of the Mn 3d orbitals, and can be used as a 257 fingerprint for transition metal valence (Rask et al., 1987; 258 Botton et al., 1995; Nakagawa et al., 2006; Varela et al., 2009). 259 Accordingly, these two different features, the O-K edge pre-260 peak intensity and the Mn  $L_{2,3}$  ratio have been used to 261 characterize the Mn oxidation state of our samples. 262

Figures 4a and 4b show a comparison of O-K and Mn-L263EEL spectra for  $\varepsilon$ -MnO2 (in black) and LaSr-2 × 4 nanowires264(in red), which show evident differences between these two265phases. For the LaSr-2 × 4 nanowires, the O-K edge pre-peak266(peak A) is significantly suppressed and shifted to higher267



**Figure 4.** a and b: O-K edge and Mn-L edge averaged electron energy loss (EEL) spectra of a  $\varepsilon$ -MnO<sub>2</sub> nanowire (in black) and a LaSr-2 × 4 nanowire (in red), respectively.



**Figure 5. a**: Left panel: High resolution Z-contrast image of a LaSr-2×4 nanowire along the [100] zone axis. A LaSr-2×4 nanowire cell model has been highlighted on the Z-contrast image, with yellow circles marking the Sr columns position, blue circles the La column position, and red and green circles the O and Mn positions, respectively. Right panel: EELS data acquired in a line scan showing the evolution along different planes of the LaSr-2×4 nanowire of the O-K, Mn-L and La-M edges along the red arrow on **a**. Red arrow indicates the direction of the scan. **b**: From top to bottom, annular dark field (ADF) simultaneous signal intensity, normalized pre-peak intensity of the O-K edge and  $L_{2,3}$  ratio profiles along the direction of the red arrow in panel **a**. Positions 1 and 2 mark two different Mn sites. **c**: A drawing of the model of the LaSr-2×4 nanowire unit cell along [010] (Carretero-Genevrier et al., 2011), with the relative direction of the electron beam during the scanning.

energies compared with the center peak (peak B). Both the 268 intensity and the distance ( $\Delta E$ ) between peak A and B are an 269 accurate indicator of valence change (Luo et al., 2009; Varela 270 et al., 2009). Also, analysis of the Mn  $L_{2,3}$  edge confirms the 271 change of manganese oxidation state. The resulting  $L_{2,3}$ 272 ratios have values of 2 and 3.2 for  $\varepsilon$ -MnO<sub>2</sub> and LaSr-2×4 273 nanowires, respectively. According to previous reports, the 274 275  $L_{2,3}$  ratio increases with the oxidation of manganite perovskites (Varela et al., 2009). Hence, doping the  $\varepsilon$ -MnO<sub>2</sub> 276 nanowires with La and Sr cations induces a mixed-valence 277

 $Mn^{3+}/Mn^{4+}$  as in perovskite manganites, This mixed 278 valence state may be associated with double exchange and 279 result in long range ferromagnetism (Dagotto et al., 2001). 280 Also, the particular atomic arrangement of the  $Mn^{3+}/Mn^{4+}$  281 manganese atoms within the monoclinic structure may 282 change the interatomic distances and modify the exchange 283 interaction stabilizing a ferromagnetic configuration. 284

Further evidence of the mixed-valence state of the 285 LaSr-2 × 4 comes from spectrum imaging. Figure 5 displays a 286 set of data used to extract the local variation of the Mn

oxidation state. Figure 5a shows a high-resolution Z-contrast 287 image of the LaSr- $2 \times 4$  nanowire along its [100] zone axis, 288 together with a line scan along the [001] axis. The ripples in 289 the O, La, and Mn signals along the scan line are a reflection 290 of the positions of the elements. A model structure, overlaid 291 on the Z-contrast image, was proposed in a previous work 292 from a similar line scan (Carretero-Genevrier et al., 2011). 293 Figure 5c shows the proposed structure of the LaSr-2  $\times$  4 294 nanowires viewed along its *b*-axis and its relative orientation 295 with respect to the electron beam during the EEL scan. In 296 this work, we have used a line scan with a longer acquisition 297 time, 1 s per pixel, to probe the changes in the fine structure 298 at different atomic positions. We have used the normalized 299 O-K pre-peak intensity (compared with the center O-K edge 300 peak, B) and the Mn  $L_{2,3}$  ratio to characterize our sample 301 (Fig. 4). The spatial modulation observed on both signals is 302 the fingerprint for a local variation of the Mn oxidation state. 303 When the scanning probe steps on the Mn column that 304 has La cations as first neighbors (position 1 in Fig. 5b) the 305 normalized pre-peak intensity decreases while the  $L_{2,3}$  ratio 306 increases. On the other hand, when the probe steps on the 307 Sr/Mn column (position 2 in Figure Fig. 5b) the normalized 308 pre-peak intensity increases and the  $L_{2,3}$  ratio decreases. 309 Comparing these results with those obtained from the 310  $\varepsilon$ -MnO<sub>2</sub> phase (Fig. 4) one can state that the Mn cations 311 surrounded by La (position 1) have a lower oxidation com-312 pared with those found in position 2. 313

#### SUMMARY AND CONCLUSIONS 314

In summary, using EMCD we have studied the magnetism of 315 a single LaSr-2  $\times$  4 nanowire at room temperature. The 316 nanowires are ferromagnetic and EMCD measurements 317 show that there is a significant orbital component to the 318 magnetic moment and that it is aligned antiparallel to the 319 spin moment. These results suggest a strong spin orbit cou-320 pling, since oppositely oriented dipole moments have lower 321 energy than those aligned parallel with each other. Our 322 finding suggests, as expected from the third Hund rule, that 323 Mn shells are less than half-filled, and that the origin of 324 ferromagnetism may reside in a double-exchange-like 325 mechanism. Indeed, spatially resolved EELS confirms the 326 presence of mixed-valence Mn cations at different sites as a 327 result of the ordered arrangement of the La<sup>3+</sup> and Sr<sup>2+</sup> 328 cations within the structure. However, the electronic struc-329 ture of these monoclinic LaSr-2×4 nanowires differs from 330 its perovskite-like counterparts and the fine structure of the 331 O-K edge presents significant changes compared with man-332 ganites (Luo et al., 2009; Varela et al., 2009). Although the 333 pre-peak in the O-K edge of manganites measures the elec-334 tronic occupancy of the hybridized O 2p and Mn 3d states, 335 and hence the oxidation state of the Mn, the lattice effects 336 must play a key role in the structure of the O-K edge in the 337 case of the LaSr-2×4 nanowires. The different disposition 338 and arrangement of La and Sr cations in the new structure of 339 the nanowires may well affect the Mn-O bonds of MnO<sub>6</sub> 340 octahedra. While further work is needed on the theoretical 341

side to interpret the features of the electronic structure of our 342 nanowires, our results pave the way toward harnessing the 343 strong magnetism at room temperature measured in LaSr-344  $2 \times 4$  manganese oxide molecular sieve nanowires. 345

### ACKNOWLEDGMENTS

We acknowledge the financial support from MICINN 347 (MAT2008-01022, Consolider NANOSELECT CSD 2007-348 00042). We thank Etienne Ferain and it4ip company for 349 supplying the nanoporous polymer membranes. Work at 350 ORNL was supported by the Office of Science, Materials 351 Sciences and Engineering Division of the U.S. Department of 352 Energy (M.V.). A.C.G. acknowledges the financial support 353 from Collège de France foundation, LCMCP for his Visiting 354 Scientist position and INL-CNRS for his detachment. We 355 thank David Montero and IMPC for use of the FEG-SEM 356 facilities. J.G. is thankful for financial support from the 357 European Research Council Starting Investigator Award, 358 grant no. 239739 STEMOX and from the Spanish Govern-359 ment (RyC-2012-11709). M.G. Acknowledges the Spanish 360 Government (RyC-2009-04335 contract). 361

### References

- BOTTON, G.A., APPEL, C.C., HORSEWELL, A. & STOBBS, W.M. (1995). 363 Quantification of the EELS near-edge structures to study Mn 364 doping in oxides. J Microsc 180, 211-216. 365
- CALMELS, L., HOUDELLIER, F., WAROT-FONROSE, B., GATEL, C., 366 HÿTCH, M., SERIN, V., SNOECK, E. & SCHATTSCHNEIDER, P. (2007). 367 Experimental application of sum rules for electron energy loss 368 magnetic chiral dichroism. Phys Rev B 76, 060409. 369
- CARRETERO-GENEVRIER, A., GÁZQUEZ, J., IDROBO, J.C., ORÓ, J., 370 Arbiol, J., Varela, M., Ferain, E., Rodríguez-Carvajal, J., 371 PUIG, T., MESTRES, N. & OBRADORS, X. (2011). Single crystalline 372 La<sub>0.7</sub>Sr<sub>0.3</sub>MnO<sub>3</sub> molecular sieve nanowires with high 373 temperature ferromagnetism. J Am Chem Soc 133, 4053-4061. 374
- CARRETERO-GENEVRIER, A., GAZQUEZ, J., MAGÉN, C., VARELA, M., 375 Ferain, E., Puig, T., Mestres, N. & Obradors, X. (2012). 376 Chemical synthesis of oriented ferromagnetic LaSr-2×4 377 manganese oxide molecular sieve nanowires. Chem Commun 378 (Camb) 48, 6223-6225. 379
- CARRETERO-GENEVRIER, A., GÁZQUEZ, J., PUIG, T., MESTRES, N., 380 SANDIUMENGE, F., OBRADORS, X. & FERAIN, E. (2010). Vertical 381 (La,Sr)MnO<sub>3</sub> nanorods from track-etched polymers directly 382 buffering substrates. Adv Functional Mater 20, 892-897. 383
- CARRETERO-GENEVRIER, A., GICH, M., PICAS, L., GAZQUEZ, J., 384 DRISKO, G.L., BOISSIERE, C., GROSSO, D., RODRIGUEZ-CARVAJAL, J. 385 & SANCHEZ, C. (2013). Soft-chemistry-based routes to epitaxial 386  $\alpha$ -quartz thin films with tunable textures. *Science* **340**, 827–831. 387
- CARRETERO-GENEVRIER, A., MESTRES, N., PUIG, T., HASSINI, A., ORÓ, J., 388 Pomar, A., Sandiumenge, F., Obradors, X. & Ferain, E. (2008). 389 Single-crystalline La<sub>0.7</sub>Sr<sub>0.3</sub>MnO<sub>3</sub> nanowires by polymer-390 template-directed chemical solution synthesis. Adv Mater 20, 391 3672-3677. 392
- CARRETERO-GENEVRIER, A., PUIG, T., OBRADORS, X. & MESTRES, N. 393 (2013). Ferromagnetic 1D oxide nanostructures grown from 394 chemical solutions in confined geometries. Chem Soc Rev 395 doi:10.1039/60288e. 396

346

362

- 397 CHEN, C.T.T., IDZERDA, Y.U.U., LIN, H.J.J., SMITH, N.V.V., MEIGS, G.,
- 398 Chaban, E., Ho, G.H.H., Pellegrin, E. & Sette, F. (1995).
- Experimental confirmation of the X-ray magnetic circular dichroism sum rules for iron and cobalt. *Phys Rev Lett* 75,
- 401 152–155.
- 402 DAGOTTO, E., HOTTA, T. & MOREO, A. (2001). Colossal
  403 magnetoresistant materials: The key role of phase separation.
  404 *Phys Rep* 344, 1–153.
- LIDBAUM, H., RUSZ, J., RUBINO, S., LIEBIG, A., HJÖRVARSSON, B.,
  OPPENEER, P.M., ERIKSSON, O. & LEIFER, K. (2010). Reciprocal and
  real space maps for EMCD experiments. *Ultramicroscopy* 11, 1380–1389.
- 409 Luo, W., VARELA, M., TAO, J., PENNYCOOK, S.J. & PANTELIDES, S.T.
- Q410 (2009). Electronic and crystal-field effects in the fine structure of
  411 electron energy-loss spectra of manganites. *Phys Rev B* 79,
  412 52405.
- AI3 NAKAGAWA, N., HWANG, H.Y. & MULLER, D.A. (2006). Why some
  interfaces cannot be sharp. *Nat Mater* 5, 204–209.
- RASK, J.H., MINER, B.A. & BUSECK, P.R. (1987). Determination of
  manganese oxidation states in solids by electron energy-loss
  spectroscopy. *Ultramicroscopy* 21, 321–326.
- 418 RZIHA, T., GIES, H. & RIUS, J. (1996). RUB-7, a new synthetic
  419 manganese oxide structure type with a 2×4 tunnel. *Eur J*420 *Mineral* 8, 675–686.
- 421 SALAFRANCA, J., GAZQUEZ, J., PÉREZ, N., LABARTA, A., PANTELIDES, S.T.,
- 422 PENNYCOOK, S.J., BATLLE, X. & VARELA, M. (2012). Surfactant 423 organic molecules restore magnetism in metal-oxide
- nanoparticle surfaces. *Nano Lett* **12**, 2499–2503.
- 453

- SCHATTSCHNEIDER, P., RUBINO, S., HÉBERT, C., RUSZ, J., KUNES, J.,
  NOVÁK, P., CARLINO, E., FABRIZIOLI, M., PANACCIONE, G. & ROSSI, G.
  (2006). Detection of magnetic circular dichroism using a
  transmission electron microscope. *Nature* 441, 486–488.
  428
- SCHATTSCHNEIDER, P., STÖGER-POLLACH, M., RUBINO, S., SPERL, M.,
  HURM, C., ZWECK, J. & RUSZ, J. (2008). Detection of magnetic
  circular dichroism on the two-nanometer scale. *Phys Rev B* 78,
  104413.
- VARELA, M., FINDLAY, S., LUPINI, A., CHRISTEN, H., BORISEVICH, A., 433
  DELLBY, N., KRIVANEK, O., NELLIST, P., OXLEY, M., ALLEN, L. & 434
  PENNYCOOK, S. (2004). Spectroscopic imaging of single atoms 435
  within a bulk solid. *Phys Rev Lett* 92, 095502. 436
- VARELA, M., LUPINI, A.R., BENTHEM, K., VAN BORISEVICH, A.Y., 437
  CHISHOLM, M.F.F., SHIBATA, N., ABE, E., PENNYCOOK, S.J.J., LUPINI, 438
  A.R. & BORISEVICH, A.Y. (2005). Materials characterization in the 439
  aberration-corrected scanning transmission electron 440
  microscope. Ann Rev Mater Res 35, 539–569. 441
- VARELA, M., OXLEY, M., LUO, W., TAO, J., WATANABE, M., LUPINI, A.,
  PANTELIDES, S. & PENNYCOOK, S. (2009). Atomic-resolution
  imaging of oxidation states in manganites. *Phys Rev B* 79, 444
  085117. 445
- ZHANG, Z.H., TAO, H.L., HE, M. & LI, Q. (2011). Origination of
   electron magnetic chiral dichroism in cobalt-doped ZnO dilute
   magnetic semiconductors. Scr Mater 65, 367–370.
- ZHANG, Z.H., WANG, X., XU, J.B., MULLER, S., RONNING, C. & LI, Q. 449 (2009). Evidence of intrinsic ferromagnetism in individual dilute magnetic semiconducting nanostructures. *Nat Nanotech* 4, 451 523–527. 452